

1 **Wintertime CO₂ Fluxes in an Arctic Polynya Using**
2 **Eddy Covariance: Evidence for Enhanced Air–Sea**
3 **Gas Transfer During Ice Formation**

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Abstract.

Between Nov. 1 2007 and Jan. 31 2008, we calculated the air-sea flux of CO₂, sensible heat, and water vapour in an Arctic polynya system (Amundsen Gulf, Canada) using eddy covariance equipment deployed on the research icebreaker *CCGS Amundsen*. During this time period, Amundsen Gulf was a dynamic sea ice environment composed primarily of first year ice with open water coverage varying between 1–14%. In all cases where measurements were influenced by open water we measured CO₂ fluxes that were 1–2 orders of magnitude higher than those expected under similar conditions in the open ocean. Fluxes were typically directed towards the water surface with a mean flux of $-4.88 \mu\text{mol m}^{-2} \text{s}^{-1}$ and a maximum of $-27.95 \mu\text{mol m}^{-2} \text{s}^{-1}$. One case of rapid outgassing (mean value $+2.10 \mu\text{mol m}^{-2} \text{s}^{-1}$) was also observed. The consistent pattern of enhanced gas exchange over open water allows us to hypothesize that high water-side turbulence is the main cause of these events. Modification of the physical and chemical properties of the surface seawater by cooling and brine rejection may also play a role. A rough calculation using an estimate of open water coverage suggests that the contribution of these events to the annual regional air-sea CO₂ exchange budget may make the winter months as important as the open water months. Although high, the uptake of CO₂ fits within mixed layer dissolved inorganic carbon budgets derived for the region by other investigators.

1. Introduction

25 In order to properly forecast the effects of climate change, general circulation models
26 need to adequately account for sources and sinks of CO₂. The global marine system plays
27 a major role in cycling CO₂ and presently absorbs about 2.2 PgC year⁻¹ [Denman *et al.*,
28 2007], which offsets about 30% of present anthropogenic emissions. However, the rate of
29 CO₂ uptake is not consistent across all oceans. On an annual basis a given region may
30 behave anywhere on the spectrum from a strong source of CO₂ to a strong sink, and
31 significant inter- and intra-annual variability may also exist [Takahashi *et al.*, 2009]. This
32 spatio-temporal variability arises from variability in the processes controlling CO₂ fluxes.

33 For the open ocean, research has advanced to the point where these processes are known
34 well enough to make reasonable flux estimates at a wide range of scales (see review by
35 Wanninkhof *et al.* [2009]). Typically, estimates of CO₂ flux (F_{CO_2}) are computed using a
36 form of the bulk flux equation:

$$F_{CO_2} = \alpha k (pCO_{2sw} - pCO_{2air}) \quad (1)$$

37 where α is the solubility of CO₂ in water, pCO_{2sw} is the partial pressure of CO₂ in the
38 surface seawater, pCO_{2air} is the partial pressure of CO₂ in the atmosphere and k is the gas
39 transfer velocity. Using this approach, the air-sea gradient of CO₂ ($pCO_{2sw} - pCO_{2air}$,
40 commonly denoted ΔpCO_2) determines the potential for exchange, while the transfer
41 velocity encompasses the processes that control the rate at which the exchange can occur.
42 The main determinant of transfer velocity is water-side turbulence, which itself is mainly
43 determined by wind velocity through its relationship with momentum flux [Jähne, 1987].

44 Many other factors influence water-side turbulence, such as wave state [*Bock et al.*, 1999;
45 *Zappa et al.*, 2004], surface films [*Jähne*, 1987; *Frew et al.*, 2004; *Frew*, 1997], rain [*Ho*
46 *et al.*, 2004; *Takagaki and Komori*, 2007; *Zappa et al.*, 2009], tides [*Zappa et al.*, 2007],
47 and buoyancy [*McGillis et al.*, 2004]. In addition, several processes not directly related to
48 turbulence also affect transfer velocity, such as chemical enhancement [*Bolin*, 1960; *Kuss*
49 *and Schneider*, 2004] and bubbles from breaking waves [*Asher et al.*, 1996; *Wolf*, 1997;
50 *Wolf et al.*, 2007]. Despite the myriad processes affecting gas exchange, wind velocity
51 alone is typically used to estimate transfer velocity in the open ocean with mature wave
52 fields [*Wanninkhof et al.*, 2009]. As such, numerous parameterizations to estimate k
53 from wind speed have been created based on tank experiments [*Liss and Merlivat*, 1986],
54 modelling exercises [*Wanninkhof*, 1992; *Sweeney et al.*, 2007], and field studies conducted
55 primarily at low and mid-latitudes [*Ho et al.*, 2006; *Nightingale et al.*, 2000; *Wanninkhof*
56 *and McGillis*, 1999].

57 At high latitudes (e.g. the Arctic), the processes that control CO₂ fluxes are not well
58 known. Depending on the season and location, a given region of the Arctic Ocean may
59 be ice free or it may be covered by sea ice of variable concentration, thickness and ther-
60 modynamic state. During the open water season it is reasonable to assume that what we
61 understand about open-ocean fluxes would be applicable, but as soon as sea ice is present
62 existing parameterizations of transfer velocity are likely invalid. Although sea ice is per-
63 meable to gas exchange under certain conditions [*Gosink et al.*, 1976], the mechanisms
64 that control the rate of exchange are very different from the open ocean. Furthermore,
65 the open water that does remain in an icescape experiences different controls on near-
66 surface turbulence; fetch limitations [*Wolf*, 2005] imposed by surrounding ice floes and

67 the generation of turbulence due to ice formation [*McPhee and Stanton*, 1996] are two
68 examples of those unique controls.

69 The initial freeze-up and growth of sea ice has generated considerable interest, since the
70 process significantly modifies the chemistry of the surface ocean and because dissolved
71 inorganic carbon (DIC) may be driven down from the surface with rejected brines in
72 what has been termed a sea ice CO₂ pump [*Rysgaard et al.*, 2007, 2009; *Anderson et al.*,
73 2004] . A water column study by *Anderson et al.* [2004] in Svalbard found high DIC and
74 elevated chlorofluorocarbon levels in deep waters, which they hypothesized originated from
75 enhanced air-sea exchange of CO₂ during ice formation. Some support for this enhanced
76 exchange was recently presented in a tank study by *Loose et al.* [2009]. In this paper, we
77 describe the first eddy covariance observations of such flux enhancements over a natural
78 sea ice surface.

2. Study Area

79 The data presented in this paper were collected between Nov. 1, 2007 and Jan. 31,
80 2008 during the International Polar Year Circumpolar Flaw Lead System Study (CFL) in
81 Amundsen Gulf and the southeaster Beaufort Sea (Figure 1). The region is subject to a
82 complex annual ice cycle which has been summarized by *Galley et al.* [2008]. The open
83 water season (defined as sea ice concentration $\leq 20\%$) typically lasts 10 weeks, starting
84 in late July. Freeze-up occurs in early October and is characterized by initial landfast
85 ice growth along the coastal margins. The ice which forms offshore in Amundsen Gulf
86 typically remains mobile during the time period of this study (shaded areas in Figure 1),
87 creating an icescape which is characterized by small transient leads and polynyas. Later in
88 the winter the eastern half of Amundsen Gulf may become landfast, and on some occasions

89 the western portion becomes landfast as well. The Beaufort Sea pack ice remains mobile
90 throughout the winter, rotating with the predominant Beaufort gyre to create persistent
91 linear flaw lead features (Figure 1). The mean spring breakup for Amundsen Gulf is
92 early June, which creates the feature commonly referred to as the Cape Bathurst polynya
93 (Figure 1) which in some years extends well into eastern Amundsen Gulf.

94 Observations have shown that the region experiences significant air-sea $p\text{CO}_2$ gradients
95 in the fall. *Mucci et al.* [2010] observed $\Delta p\text{CO}_2$ ranging from -138 to -28 μatm from Sep.–
96 Nov. 2003, and *Murata and Takizawa* [2003] observed gradients of similar magnitude
97 during three years of cruises in Aug.–Sep., 1998–2000. Observations made during the
98 CFL study showed that significant undersaturation ($\Delta p\text{CO}_2$ typically around -70 μatm)
99 persisted through the end of January 2008 in offshore Amundsen Gulf [*Shadwick et al.*,
100 2011].

101 During the winter season, the persistent flaw leads and polynyas in combination with
102 strong local $p\text{CO}_2$ gradients make this study area an ideal location for examining the effect
103 of freezing sea ice on gas exchange.

3. Methods

3.1. Atmospheric Instrumentation

104 For the duration of the experiment a guyed open-lattice tower at the bow of the ship
105 was instrumented with eddy covariance and meteorological equipment. The flux instru-
106 mentation consisted of a Gill Windmaster Pro sonic anemometer/thermometer, a LI-COR
107 LI-7500 open path CO₂/H₂O gas analyzer and a Systron Donner MotionPak. The flux
108 instrumentation was located at a height of 14 m above the surface (7 m above the deck

109 of the ship), with the exception of the MotionPak which was located at the midpoint of
110 the tower.

111 The meteorological equipment consisted of a conventional anemometer for wind speed
112 and direction (RM Young 05103, height=15 m), a temperature/relative humidity probe
113 (Vaisala HMP45C212, height=14 m) and a pressure sensor (RM Young 61205V). An
114 array of radiation sensors was deployed on top of the wheelhouse of the ship, consisting of
115 a photosynthetically active radiation (PAR) sensor (Kipp & Zonen PARlite), an incoming
116 shortwave radiation sensor (Eppley PSP) and an incoming longwave radiation sensor
117 (Eppley PIR).

3.2. Surface Water $p\text{CO}_2$ Instrumentation

118 Surface water from a dedicated scientific intake line (depth ~ 5 m) was continuously
119 sampled for $p\text{CO}_{2sw}$ using a shower-type equilibrator which cycled headspace air through
120 a LI-COR LI-7000 CO₂/H₂O gas analyzer [Körtzinger *et al.*, 1996]. The gas analyzer was
121 calibrated daily using ultra-high purity N₂ as a zero gas and a CO₂/air mixture traceable to
122 WMO standards as a span gas. The instrument was located in the engine room very close
123 to the water intake, but a slight warming of the sample water relative to results from CTD
124 casts was detected by a thermocouple in the equilibrator. This warming effect was very
125 consistent, allowing correction of $p\text{CO}_{2sw}$ for thermodynamic effects following *Takahashi*
126 *et al.* [1993]. After correction, the $p\text{CO}_{2sw}$ measurements showed good agreement (r^2
127 = 0.9, mean difference = 19 μatm , no statistically significant bias) with independent
128 calculations from DIC/TA measurements (see *Shadwick et al.* [2011] for a description of
129 the DIC/TA dataset and methods).

3.3. Study Design

130 The CFL study was a unique over-wintering experiment because the research vessel
131 remained mobile through the entire winter. The goal of this strategy was to create a
132 time series of the seasonal evolution of the flaw lead/polynya system. Logistically, this
133 meant that the specific location and operation of the vessel was highly opportunistic;
134 when ice conditions allowed ship to move freely, spatial sampling was conducted, but
135 when ice conditions were more severe the ship was positioned in large consolidated floes
136 and allowed to drift. These floes were typically occupied for 1 – 7 days, depending on the
137 stability of the floe and whether or not it was drifting outside of the study area. When
138 repositioning was necessary, the ship would break out of the floe and either break ice or
139 transit through small flaw leads until a more suitable floe was located.

3.4. Eddy Covariance

140 The study design allowed us to examine a sea ice system which would otherwise be
141 inaccessible, but it does have implications for the eddy covariance technique which is best
142 suited for a stationary tower over a homogenous surface. To help address these issues, we
143 filtered the data to ensure that each eddy covariance run was not subject to significant
144 changes either in ship operation or atmospheric conditions. If the ship was under power,
145 ship velocity and course over ground were required to be consistent (within ± 3.7 km
146 hr^{-1} of mean for velocity and $\pm 27.5^\circ$ of mean for course). Relative wind direction was
147 also required to be consistent within $\pm 27.5^\circ$ of the mean, and it was further restricted
148 to within $\pm 90^\circ$ of the bow of the ship to reduce the effects of flow distortion. To help
149 with the issue of non-homogeneous surfaces, we found it useful to break the data up into
150 individual case studies during time periods where flux data collection was consistent and

151 the ship location, atmospheric conditions and sea ice conditions were fairly uniform (see
 152 Table 1, Figure 2).

153 Filtering was also necessary to remove instances where atmospheric conditions nega-
 154 tively impacted the flux instruments. The LI-7500 outputs a diagnostic value that warns
 155 of lens obstruction, which during this study was most often caused by accretion of rime.
 156 We filtered out all instances where the diagnostic value exceeded its normal operating
 157 range, creating a fairly significant loss of data. The sonic anemometer was less influenced
 158 by riming, but filtering was carried out based on the characteristically erratic performance
 159 of the instrument that occurs under such circumstances.

160 The LI-7500 used in this study makes high frequency (10Hz) measurements of the molar
 161 concentrations of CO₂ and water vapour (c_{co_2} and c_v respectively). By combining these
 162 measurements with high frequency vertical wind velocity (w) measurements from the
 163 sonic anemometer, the flux of CO₂ is calculated over an averaging period (in this case, 30
 164 minutes) via:

$$F_c = \overline{w'c'_{co_2}} + \frac{\bar{c}_{co_2}}{\bar{c}_d} \left[\overline{w'c'_v} + \bar{c}_a \frac{\overline{w'T'}}{\bar{T}} \right] \quad (2)$$

165 where the overbars denote averaged quantities, the primes indicate fluctuations around
 166 a mean value, T is air temperature, c_d is the dry air molar concentration, and c_a is the
 167 moist air molar concentration [Leuning, 2004]. The second term on the right hand side
 168 of equation 2 is the so-called WPL correction (or dilution correction) that must be used
 169 for open path sensors [Webb *et al.*, 1980]. The necessary high frequency T measurements
 170 are determined from sonic temperature (measured by the sonic anemometer), which were
 171 converted to T following *Kaimal and Gaynor* [1991]. The MotionPak provides 3-axis

172 measurements of acceleration and angular velocity which were used to correct w for ship
173 motion. The techniques for this correction were first adapted for ships by *Mitsuta and*
174 *Fujitani* [1974], and later refined by other investigators [*Fujitani*, 1981; *Dugan et al.*, 1991;
175 *Anctil et al.*, 1994; *Edson et al.*, 1998].

176 The utility of open path sensors for measuring CO₂ fluxes has recently been debated
177 for conditions where low fluxes are expected [*Burba et al.*, 2008; *Amiro*, 2010; *Ono et al.*,
178 2008] and in the marine environment [*Prytherch et al.*, 2010]. During the non-growing
179 season over several terrestrial ecosystems, significant uptakes of CO₂ have been observed
180 and identified as artifacts of the LI-7500 gas analyzer [*Amiro*, 2010; *Ono et al.*, 2008;
181 *Hirata et al.*, 2007]. Work by *Burba et al.* [2008] have shown that a heat flux generated
182 by the electronics of the LI-7500 is likely the most significant contributor to this bias,
183 especially at low air temperatures. Other suggestions include pressure fluctuations at
184 high wind velocities that are not usually included in the WPL correction [*Järvi et al.*,
185 2009] and incomplete WPL corrections due to poor energy balance closure [*Ono et al.*,
186 2008]. The magnitude of these discrepancies are usually on the order of $1 \mu\text{mol m}^{-2} \text{s}^{-1}$,
187 which in most terrestrial systems during the growing season is a small percentage of the
188 total flux. However, typical magnitudes of CO₂ flux in the open ocean are less than 1
189 $\mu\text{mol m}^{-2} \text{s}^{-1}$ (e.g., *McGillis et al.* [2001]).

190 A further difficulty of working with an open path analyzer in a marine environment is
191 an apparent sensitivity to contamination of the sensor lens by impurities (most likely salt
192 particles) [*Kohsiek*, 2000; *Prytherch et al.*, 2010]. The contamination appears to cause a
193 portion of water vapour fluctuations to be mis-recorded as fluctuations of CO₂ (an effect

194 known as “crosstalk”), and can lead to CO₂ fluxes an order of magnitude higher than
195 expected [*Prytherch et al.*, 2010].

196 Corrections have been proposed for both the sensor heating and H₂O crosstalk issues.
197 *Burba et al.* [2008] proposed several ways in which the heat flux of the LI-7500 can be
198 estimated and added to the WPL correction. In this study, we have adopted their multi-
199 variate regression model for determining the sensor heat flux from air temperature, wind
200 velocity and incoming longwave/shortwave radiation. *Prytherch et al.* [2010] proposed
201 a correction for the H₂O crosstalk (termed the “PKT” correction) in which an iterative
202 approach is used to remove unwanted correlation between the CO₂ and H₂O signals. This
203 correction has also been adopted for this study, but as we discuss in section 5.1 we found
204 it to be unreliable. Thus, all CO₂ flux values reported herein include only the *Burba et al.*
205 [2008] correction along with the usual WPL corrections.

3.5. RADARSAT-1 Imagery

206 To aid in the identification of ice conditions and to quantify the amount of open water
207 within the study area over the period, fourteen (14) RADARSAT-1 ScanSAR narrow beam
208 images acquired between Nov. 6 2007 and Jan. 28 2008 were classified. RADARSAT-1
209 ScanSAR narrow beam mode has a resolution of 50 m and a nominal coverage area of 300
210 x 300 km. Each of the images were geo-referenced and calibrated to σ° , then geographi-
211 cally cropped using latitudinal bounds 70° and 71.5°N and longitudinal bounds 122° and
212 126°W. The calibrated, geo-referenced sub-images were then subjected to a median fil-
213 ter with a 3x3 window size to reduce the “speckle” noise common to synthetic aperture
214 RADAR (SAR) imagery while preserving edges. Edge preservation is very important
215 when linear features such as leads are the predominant form of open water at this time of

216 year. Finally, each sub-image was manually classified according to the principles set forth
217 by the Canadian Ice Service (CIS) SAR ice interpretation guide [*Canadian Ice Service*,
218 unpublished].

4. Results

4.1. Observations of High CO₂ Flux Events

219 4.1.1. Case 1: Nov. 2 04:30 - Nov. 3 09:30

220 On Nov. 2, the ship conducted a transect across the mouth of Amundsen Gulf. A
221 RADARSAT-1 image was acquired on Nov. 2 at 01:54 (all times herein reported as UTC)
222 just prior to the start of the transect, which clearly shows that the region was a mix
223 of open water, old ice floes, and newly forming grease ice (Figure 3). Due to the ice in
224 the area the intake line for the $p\text{CO}_{2sw}$ system was clogged so that we could not obtain
225 measurements of $\Delta p\text{CO}_2$. However, samples collected in the region in the previous two
226 days showed that the $\Delta p\text{CO}_2$ was around $-80 \mu\text{atm}$.

227 Over this time period, we measured a flux of up to $-4.26 \mu\text{mol m}^{-2} \text{s}^{-1}$, with a mean
228 value of $-1.81 \mu\text{mol m}^{-2} \text{s}^{-1}$. Associated with this strong CO₂ uptake was a high sensible
229 heat flux from the ocean to the atmosphere (up to 100 W m^{-2} , mean of 43.3 W m^{-2} , Figure
230 4b). This heat flux must have been driven by open water, over which a strong temperature
231 gradient forms due to the relative warmth of the ocean. We therefore interpret the strong
232 CO₂ fluxes to likewise be a signal of open water gas exchange.

233 4.1.2. Case 3: Nov. 20 01:30 – 14:45

234 The second instance where we observed particularly high CO₂ fluxes was on Nov. 20,
235 near the southern tip of Banks Island. At this point ice concentration in Amundsen Gulf
236 was very high, and the ship was parked in a 36 cm thick ice floe. Prior to this time, a

237 strong wind event from Nov. 16-17 (wind velocities peaking at about 24 m s⁻¹) created
238 significant ice motion and fracture in the region. This was followed by very low winds
239 (about 5 m s⁻¹), allowing the open water features to appear obviously on a RADARSAT-
240 1 image acquired on Nov. 20 at 01:29 as dark features (Figure 5). Early on Nov. 20,
241 easterly winds picked up quickly to about 13 m s⁻¹ and persisted through the sample case
242 (Figure 6c). This wind induced significant ice motion (the ship drifted at a mean velocity
243 of 0.6 km hr⁻¹, increasing steadily from 0.4 to 1.1 km hr⁻¹), which would have expanded
244 the open water leads. A signal of open-water fluxes was clearly evident in the heat flux
245 measurements, reaching nearly +100 W m⁻² (Figure 6b) with a mean value of +53.8 W
246 m⁻² (Table 1).

247 During this case, rapid outgassing of CO₂ was observed at a mean rate of +2.10 μmol
248 m⁻² s⁻¹. The outgassing was somewhat surprising given that the *p*CO<sub>2_{sw} system recorded
249 undersaturation of -75 μatm (Table 1), but the surface water upwind of the ship may have
250 been supersaturated. This result shows that at times our eddy covariance measurements
251 (which are an integrated flux from the upwind surfaces) are difficult to reconcile with the
252 *p*CO_{2_{sw} data (which measures at the same location as the tower).}</sub>

253 **4.1.3. Case 4: Nov. 20 16:00 – 18:30**

254 Case 4 is an extension of Case 3, but we split the two because the ship repositioned
255 (approximately 1.5 km north) to a new ice floe (Figure 5), and the CO₂ flux changed
256 markedly. The heat flux measurements from Case 4 were still heavily influenced by an
257 open water signal, with even higher values than in case 3 (mean and maximum values of
258 +111.4 and +140.7 W m⁻² respectively, Figure 6b). This is in agreement with ship-board
259 observations that the area was a mixture of ice and open water under strong wind forcings.

260 After the ship repositioned, a very strong negative flux of CO₂ was observed (mean and
261 maximum values of -9.58 and -11.43 $\mu\text{mol m}^{-2} \text{s}^{-1}$ respectively, Figure 6a). This flux
262 direction is in better agreement with the $\Delta p\text{CO}_2$ gradient observed in the area, and may
263 be a result of a change in the upwind surface to lower (i.e. undersaturated) $p\text{CO}_{2sw}$ as
264 the ship moved around the southern tip of Banks Island (Figure 5).

265 4.1.4. Case 7: Dec 1 07:00 – 12:30

266 The strongest CO₂ fluxes that we measured were on Dec. 1 during a transit along the
267 southwest coast of Banks Island. Immediately prior to this transit, the ship was drifting
268 south in an ice floe under fairly high winds (mean 11.8 m s^{-1}). This drifting event made
269 up Case 6, where no strong CO₂ fluxes or heat fluxes were observed (Table 1). The ship
270 eventually broke out of this drift, and transited through an active wind-roughened flaw
271 lead. This flaw lead event was captured in a RADARSAT-1 image taken shortly after the
272 end of case 7 (Dec. 1, 14:45, Figure 7).

273 The transit was very short, and only four 30 minute samples passed our quality control
274 tests. However, all of these samples showed very high CO₂ uptake, with flux values
275 ranging from -9.33 to -27.95 $\mu\text{mol m}^{-2} \text{s}^{-1}$ (Figure 8a). Once again, these fluxes were
276 accompanied by high sensible heat fluxes indicative of open water (Figure 8b). No $p\text{CO}_{2sw}$
277 measurements were available during the transit due to a clogged intake line, but based
278 on the $\Delta p\text{CO}_2$ during the cases bracketing this one (-52.1 and -63.6 μatm for cases 6 and
279 8, respectively, Table 1), the direction of the flux appeared to be in agreement with the
280 gradient.

281 4.1.5. Case 17: Jan 24. 08:00 - Jan. 25 05:30

282 The final instance where we measured unusually strong CO₂ fluxes was during case 17
283 in late January. During this time, the ship was drifting in an ice floe with a thickness of
284 about 100 cm. A RADARSAT image collected at 01:33 on Jan. 24 showed considerable
285 fracturing upwind of the ship (Figure 9).

286 This case was characterized by high wind velocities (up to 19 m s⁻¹, Figure 10c) which
287 caused ice drift up to 1.4 km hr⁻¹. These strong winds and ice motion drove significant
288 open water, as observed in heat flux measurements approaching 100 W m⁻² (Figure 10b).
289 Associated with this sensible heat signal was a strong, consistent CO₂ uptake (Figure
290 10a) with a mean flux of -3.15 μmol m⁻² s⁻¹, in agreement in direction with an observed
291 Δ*p*CO₂ gradient of -38.5 μatm.

4.2. Observations of Low CO₂ Fluxes

292 4.2.1. Observations in Land Fast Ice

293 Flux measurements were made in land fast sea ice on three occasions: case 2 (Nov. 8
294 02:15 – Nov. 9 00:50, 69.50 °N/123.93 °W), case 8 (Dec. 1 13:45 – Dec. 2 02:45, 71.90
295 °N/125.44 °W), and case 11 (Dec. 19 23:15 – 18:15, 71.91 °N/125.43 °W). The minimum
296 ice thickness for all of these samples was an estimated 40 cm (case 2), and the ice was
297 much thicker (approaching 100 cm) in the other two cases. Sensible heat flux in all cases
298 was small (Table 1), and in all three cases the wind direction was such that the upwind
299 fetch was composed of fast ice. This suggests that what we were measuring was indeed a
300 land fast ice signal. In these cases, mean fluxes were between +0.23 – +0.42 μmol m⁻²
301 s⁻¹ (Table 1 and Figure 2). If these measurements are reliable, they would suggest a flux
302 of CO₂ at a climatologically significant rate. However, we will show in section 5.1 that

303 these low fluxes likely cannot be distinguished from the noise and biases inherent in our
304 eddy covariance system.

305 4.2.2. Observations in High Concentration Mobile Ice

306 A second scenario in which we typically observed non-resolvable CO₂ fluxes was when
307 the ship was drifting in highly concentrated mobile sea ice. These conditions were observed
308 during case 6 (Nov. 30 05:15 – 23:00), case 9 (Dec. 2 05:30 – 22:15), case 13 (Jan. 2
309 18:15 – Jan. 6 03:30), case 14 (Jan. 10 09:15 – Jan. 11 18:45), case 15 (Jan. 13 16:00
310 – Jan. 14 00:00) and case 16 (Jan. 20 17:00 – Jan. 21 04:30). As Table 1 shows, all of
311 these cases had very low sensible heat fluxes (highest mean flux was +3.2 W m⁻², case
312 15), and mean CO₂ fluxes ranged from -0.50 – +0.58 μmol m⁻² s⁻¹ (Table 1, Figure 2).

313 4.2.3. Observations in Thin Ice

314 A final scenario which is perhaps more interesting is the observation of non-detectable
315 CO₂ fluxes in cases where other observations (field data and heat flux measurements)
316 suggest that thin ice may be present. Only case 10 (Dec. 4 23:00 – Dec. 6 12:15) falls
317 into this category. RADARSAT-1 images (not shown) acquired shortly before (Dec. 4,
318 01:21) and after this run (Dec. 7 01:33) do not show a lot of obvious thin ice, but ice
319 cores taken from the surrounding floe were only 26 cm thick. The heat flux measurements
320 (Figure 11b) were consistently positive (mean value of +15.8 W m⁻²), but lower than
321 those observed in section 4.1. Thin ice transfers heat at significant rates, but does so less
322 vigorously than open water [*Maykut*, 1978]. Although winds were moderate (5 – 7 m s⁻²,
323 Figure 11c) the Δ*p*CO₂ gradient was quite high (mean value of -86.7 μatm). If a flux
324 enhancement was occurring similar to those described in section 4.1, we would expect to
325 be able to detect it in our CO₂ flux measurements. However, Figure 11a clearly shows that

326 fluxes were not distinctly above the uncertainty inherent in the system. These findings
327 suggest that open water – not just thin ice – is required to drive CO₂ flux at the levels
328 shown in section 4.1.

5. Discussion

5.1. Sensor Uncertainties

329 The results obtained in land fast and consolidated ice (sections 4.2.1 & 4.2.2) provide an
330 opportunity to test the noise and bias inherent in our eddy covariance system. There is in
331 fact good reason to expect that CO₂ fluxes over these surfaces (thick, cold, consolidated
332 sea ice) should be zero. At surface ice temperatures below $\sim -5^{\circ}\text{C}$ and typical brine
333 salinity, the brine volume drops below 5% which inhibits liquid transport through the
334 ice [Golden *et al.*, 1998]. Loose *et al.*, [2010] examined the transport of gases near this
335 liquid transport threshold, and found the gas transfer velocity to be very small relative
336 to seawater. Similarly, Nomura *et al.*, [2006] measured small CO₂ fluxes (maximum \sim
337 $+0.01\mu\text{mol m}^{-2} \text{s}^{-1}$) over thin laboratory ice well above the liquid transport threshold. At
338 ice temperatures that reduce brine volume to below 5%, these small rates of gas exchange
339 should be effectively shut off.

340 After Nov. 28, 2007 (when most of the measurements described in sections 4.2.1 & 4.2.2
341 were made) surface ice temperatures were consistently well below -5°C and brine volumes
342 were typically below 5% [G. Carnat, unpublished data]. We would therefore expect any
343 deviation of the mean CO₂ fluxes over these surfaces from zero to be indicative of bias,
344 and any variation around that mean to be noise in the measurement system. To this end,
345 we calculated the mean and standard deviation of the raw, sensor heating corrected, and
346 crosstalk (PKT) corrected CO₂ fluxes from cases 2, 6, 8, 9, 11 & 13–16 (Table 2).

347 The uncorrected fluxes show a negative bias ($-0.45 \mu\text{mol m}^{-2} \text{s}^{-1}$), which is in the direc-
348 tion predicted by both sensor heating and water vapour crosstalk effects. The standard
349 deviation of CO₂ fluxes around this mean was $0.76 \mu\text{mol m}^{-2} \text{s}^{-1}$, indicating that noise
350 is quite high in the system. By applying only the sensor heating correction, the bias
351 moved to $+0.13 \mu\text{mol m}^{-2} \text{s}^{-1}$ – a reduction in the magnitude of the bias, but a slight
352 overcorrection. Fortunately, this correction did not add a lot of additional noise to the
353 system, as the standard deviation remained relatively unchanged ($0.77 \mu\text{mol m}^{-2} \text{s}^{-1}$).

354 The PKT correction, however, was more troublesome. We found that 55% of the sam-
355 ples from these cases produced what we determined to be an “unreasonable” correction
356 (magnitude of correction $>5.5 \mu\text{mol m}^{-2} \text{s}^{-1}$). Of the remaining samples, the net effect of
357 the correction was to actually make the CO₂ flux more negative, counter to the expected
358 direction. Furthermore, the correction added additional noise as evidenced by an increase
359 in standard deviation to $1.30 \mu\text{mol m}^{-2} \text{s}^{-1}$. Since no negative bias remains in the mean
360 flux after the sensor heating correction, we conclude that crosstalk contamination must
361 have been small for our system even prior to applying the PKT correction. Arguments
362 for a low crosstalk error in this environment have a strong physical basis, because latent
363 heat fluxes were very small (typically $<5 \text{ W m}^{-2}$) compared to the examples discussed by
364 *Prytherch et al.* [2010] ($\sim 60 \text{ W m}^{-2}$). From an eddy covariance standpoint, a low latent
365 heat flux means that water vapour is not highly correlated with vertical wind velocity,
366 and thus should not cause significant spurious correlation between CO₂ and vertical wind
367 velocity. For these reasons, we decided not to include the PKT correction in our results.

368 We propose that our system has an overall uncertainty of $\pm 0.77 \mu\text{mol m}^{-2} \text{s}^{-1}$ and a
369 bias of $+0.13 \mu\text{mol m}^{-2} \text{s}^{-1}$ based on the results of the sensor heating corrected fluxes.

370 This level of uncertainty shows that our measurements of high CO₂ flux (section 4.1) are
371 above the noise level of the system, and are not the result of a strong systematic bias.

5.2. Enhanced Gas Flux by Sea Ice Formation

372 Our results from section 4.1 indicate that in the winter mixed ice environment of the
373 Amundsen Gulf, the presence of open water drives a very rapid exchange of CO₂. For
374 comparison, under the typical $\Delta p\text{CO}_2$ ($\sim 70 \mu\text{atm}$) and wind velocity ($\sim 8 \text{ ms}^{-1}$) con-
375 ditions we encountered, the bulk flux approach (equation 1) would predict fluxes in the
376 range of -0.10 to $-0.12 \mu\text{mol m}^{-2} \text{ s}^{-1}$. Even using the maximum wind velocities observed
377 (19 ms^{-1}), we would not expect fluxes to exceed about $-1.5 \mu\text{mol m}^{-2} \text{ s}^{-1}$. Our measured
378 fluxes are therefore at times 1–2 orders of magnitude higher than what might be expected
379 under similar conditions in the open ocean.

380 Several authors have suggested that an enhancement of gas exchange due to sea ice
381 formation may exist [*Anderson et al.*, 2004; *Rysgaard et al.*, 2007; *Loose et al.*, 2009], but
382 none have described in detail the physical and chemical processes which may account for
383 it. Our study likewise lacks the necessary ancillary observations to show conclusively what
384 processes are responsible for enhanced gas transfer, but in this section we propose two
385 key hypotheses to explain it: (1) enhanced water side turbulence driven by rapid cooling
386 and brine rejection, and (2) modification of the carbonate system of the surface seawater.
387 These hypotheses are summarized in Figure 12 and discussed below.

5.2.1. Enhanced Water Side Turbulence

389 At the upwind side of a flaw lead, a significant heat flux occurs due to the exposure
390 of the relatively warm (i.e. $\sim -1.8 \text{ }^\circ\text{C}$) water to the very cold atmosphere ($\sim -10 - -25$
391 $^\circ\text{C}$). This cooling creates a destabilization of the water surface and generates buoyancy

392 fluxes that may enhance turbulence. *McGillis et al.* [2004] observed a 40% enhancement
393 of CO₂ fluxes during modest nighttime cooling (sensible heat fluxes on the order of 1-10
394 W m⁻²) in the equatorial pacific, which was attributed mostly to these buoyancy fluxes.
395 In a situation such as the one shown in Figure 12 where sensible heat fluxes are 1-2 orders
396 of magnitude higher, this enhancement is likely to be much more pronounced.

397 A second process that may drive high turbulence is the rejection of dense brines by
398 frazil ice formation. Frazil ice is small, unconsolidated ice crystals that are primarily
399 generated just below the surface [*Ushio and Wakatsuchi*, 1993]. It is easily transported
400 away from the open water site, creating a region of rapid ice formation but persistent open
401 water. Frazil ice crystals are thought to be essentially pure [*Omstedt*, 1985] which means
402 that their formation results in the rejection of any solutes, which must create density
403 instabilities and drive enhanced turbulence similar to the effect of heat loss.

404 Unfortunately, turbulence in these systems has not been well studied. Between Nov.
405 16 – Dec. 18, one of our collaborators collected 175 profiles of turbulent kinetic energy
406 dissipation rate (ϵ) from a minimum depth of 10 m using a vertical microstructure turbu-
407 lence profiler (VMP, see *Bourgault et al.* [2008] for instrument details). ϵ at 10 m reached
408 values of $O(10^{-5})$ W kg⁻¹ on a few (~ 4) profiles, with an approximately exponential
409 decrease with depth [D. Bourgault, pers. comm]. Extrapolating above 10 m suggests
410 surface dissipation rates that may have occasionally reached $O(10^{-4})$ W kg⁻¹. These
411 values are considerably higher than ϵ measured under refrozen leads at a similar depth
412 by *McPhee and Stanton* [1996] ($O(10^{-8}) - O(10^{-7})$ W kg⁻¹). The exponential shape of
413 the dissipation measurements points to surface turbulence generation, but given that the
414 dominant ice cover must restrict wind and wave action, sea ice processes (likely including

ice drift and brine rejection) must play an important role in this system. *Zappa et al.* [2007] showed that ϵ is a better predictor of gas transfer velocity than wind in systems where turbulence is generated from other sources. Our maximum predicted surface ϵ values are similar to the highest ϵ measurements made by *Zappa et al.* [2007] in coastal zones and tidal estuaries, but they are not high enough to account for the rate of gas transfer we observed. However, the VMP was almost always deployed when the ship was stationary in ice floes, and it therefore may not have captured the nature of the transient flow leads that we hypothesize to be the cause of our high observed CO₂ fluxes.

5.2.2. Modification of the Surface Seawater Carbonate System

$p\text{CO}_{2sw}$ is ultimately controlled by the equilibrium condition of the seawater carbonate system. DIC, TA, salinity and water temperature all affect this equilibrium, and thus exert a control on $p\text{CO}_{2sw}$. In terms of gas exchange, it is actually the carbonate system properties of the very thin mass diffusive layer that determines the air-sea $\Delta p\text{CO}_2$.

The most obvious modification by lead formation is cooling of the sea surface, which will reduce $p\text{CO}_{2sw}$ and increase solubility. Although the seawater will be near its freezing point, cooling beyond the freezing point (supercooling) occurs before ice formation begins. If no particles are available for the nucleation of ice crystals, supercooling can easily exceed 2° C [*Tsang and Hanley, 1985*]; a condition which can be created in the laboratory, but is not likely to exist in the Arctic. Observations of supercooling in the field are sparse, but *Skogseth et al.* [2009] observed a supercooling of ~ 0.04 °C in the bulk surface water of an open coastal polynya in Svalbard. Given that the heat loss is at the surface, this would likely translate into an even more significant cooling of the diffusive mass boundary layer – in essence, the rapid sensible heat flux would drive a very pronounced cool-skin effect.

438 This cool-skin effect would enhance uptake when the sea surface was undersaturated due
439 to increased solubility and decreased $p\text{CO}_{2sw}$, but would actually act to restrict exchange
440 when the surface was supersaturated. Given that we measured one instance of intense
441 outgassing (case 3, section 4.1.2), this process alone cannot account for the high exchange
442 rates.

443 Frazil ice formation and the accompanying rejection of brines also has the potential to
444 modify the near-surface chemistry. A decrease in solubility driven by salt rejection and
445 rising DIC/TA concentrations may either suppress or enhance gas exchange, depending
446 on the saturation state. When the sea surface is supersaturated, the added DIC/TA and
447 reduced solubility should enhance outgassing. When the sea surface is undersaturated,
448 these combined effects should suppress uptake. However, whether or not this has an in-
449 fluence on gas exchange depends on where the brines ultimately end up. On the nearby
450 Beaufort Sea Shelf, *Melling and Moore* [1995] showed that deep penetration to the pycn-
451 ocline of brine does occur at times, which suggests that modification of the near-surface
452 chemistry may not be important. *Shadwick et al.* [2011] did measure long-term surface
453 increases in salinity, and DIC/TA in Amundsen Gulf over the winter, but we do not have
454 measurements that capture the evolution of these properties on the timescale of an in-
455 dividual flaw lead event. On these short timescales, we hypothesize that the effects of
456 brine rejection will be minor. Since most of the frazil ice formation is occurring below
457 the ocean skin, there will not be much immediate modification of the chemistry of the
458 mass diffusive layer. *Ushio and Wakatsuchi* [1993] also showed that the brine rejection
459 from frazil crystals is concentrated in thin streamers that rapidly descend downwards. If
460 leads are small and short-lived, there would be a significant amount of unmodified water

461 available laterally and vertically to replace those descending brines, keeping the surface
462 water properties near-constant.

463 Ultimately, we cannot draw any firm conclusions regarding the short timescale mod-
464 ifications to the seawater carbonate system. However, given the manner in which the
465 carbonate system is entwined with many of the processes that occur with lead formation,
466 this should be a major focus of future studies.

5.3. Significance to the Amundsen Gulf Region

467 The total winter CO₂ flux through open water in the Amundsen Gulf depends on not
468 only the rate at which it occurs, but also on the areal extent of open water. In this
469 section, we combine estimates of these two variables for the purpose of computing area-
470 averaged fluxes. By calculating these fluxes, we can estimate the significance of winter
471 CO₂ exchange relative to the open water season (i.e. late spring/summer/early fall), and
472 we can determine if the fluxes are reasonable based on the water column DIC budget for
473 the region devised by *Shadwick et al.* [2011].

474 As described in section 3.5, we used RADARSAT-1 imagery to estimate the open
475 water fraction in a bounding box (122–126°W, 70–71.5°N) consistent with the one used
476 by *Shadwick et al.* [2011]. RADARSAT-1 images that captured this area were available
477 approximately every week, and the open water fraction calculated for each image are
478 shown in Figure 13. The amount of open water during the study was highly variable,
479 which probably relates to storm events in the area. To allow comparisons with the results
480 of *Shadwick et al.* [2011], monthly averages of open water fraction were calculated, and
481 are displayed in Table 3.

482 To estimate the rate of gas exchange over the area we would ideally have some way of
 483 scaling our flux measurements using $\Delta p\text{CO}_2$ and an easily-obtainable variable like wind
 484 speed, but at this point our dataset is too limited to work towards parameterization.
 485 Therefore, we simply calculated the mean uptake rate from cases 1,4,7 & 17 (those with
 486 substantial uptake; the outgassing observed during case 3 was omitted because the offshore
 487 Amundsen Gulf was undersaturated through the entire winter [*Shadwick et al.*, 2011]) to
 488 be $-4.88 \mu\text{mol m}^{-2} \text{s}^{-1}$. This rate was multiplied by the average monthly open water
 489 fraction to calculate the mean monthly fluxes shown in Table 3.

490 To address the question of whether these fluxes are reasonable we integrated the flux
 491 over each month, and calculated the net change in DIC (denoted $\Delta\text{DIC}_{as-enh}$ in Table
 492 3) that would occur assuming a 50m mixed-layer depth. *Shadwick et al.* [2011] budgeted
 493 month-by-month changes in DIC in the Amundsen Gulf mixed layer via:

$$\Delta\text{DIC}_{obs} = \Delta\text{DIC}_{bio} + \Delta\text{DIC}_{fw} + \Delta\text{DIC}_{as} + \Delta\text{DIC}_{vd} \quad (3)$$

494 where ΔDIC_{obs} was the observed monthly change in DIC, and the right hand terms are
 495 monthly changes in DIC due to biological activity (ΔDIC_{bio}), freshwater fluxes (ΔDIC_{fw}),
 496 air-sea exchange (ΔDIC_{as}) and vertical diffusion (ΔDIC_{vd}). ΔDIC_{fw} and ΔDIC_{vd} were
 497 calculated from in situ data, and ΔDIC_{as} was calculated using a bulk flux approach
 498 scaled for ice concentration. No direct method was available to measure the biological
 499 contribution, so it was calculated as a difference of the 4 other terms. With no other
 500 constraint on the biological contribution to ΔDIC_{obs} , a flux of CO₂ which is enhanced
 501 beyond ΔDIC_{as} would be mis-allocated into ΔDIC_{bio} . Thus if our calculated $\Delta\text{DIC}_{as-enh}$
 502 fits within the sum of ΔDIC_{as} and ΔDIC_{bio} it can be considered to fit in the budget.

503 Table 3 shows that this is the case in all months under consideration, except November.
504 This shows that although the fluxes associated with this enhanced air-sea exchange are
505 very high, they are not unrealistic from the standpoint of the DIC budget.

506 With respect to significance for the Amundsen Gulf region, the fluxes calculated ac-
507 counting for enhanced air-sea exchange are more than an order of magnitude higher than
508 those calculated by *Shadwick et al.* [2011] using a bulk flux approach scaled for ice con-
509 centration (Table 3). In fact, these fluxes place the air-sea exchange rates on par with the
510 open water season rates calculated by *Shadwick et al.* [2011]. This is a significant consider-
511 ation, because the typical model of a polynya's annual air-sea budget is characterized by
512 open water uptake during the autumn storm season (utilizing an initial biological $p\text{CO}_{2sw}$
513 drawdown in the spring) which is then capped by ice over the winter [*Yager et al.*, 1995].
514 The strength of annual uptake by a polynya was thought to be constrained by whether or
515 not the spring undersaturation could be utilized by open water air-sea exchange, but the
516 results from this study show that uptake may proceed beyond ice formation. It should be
517 noted, however, that not every polynya may remain undersaturated through the winter;
518 in polynyas where this is not the case, winter outgassing through open water may tip the
519 annual balance away from net uptake.

5.4. Potential Significance to the Arctic Ocean

520 As well as creating a need to re-think the seasonal evolution of gas exchange for polynyas,
521 enhanced winter gas exchange may play an important role in the broader Arctic and
522 Antarctic Oceans. *Omar et al.* [2005] used a simple extrapolation of winter air-sea CO₂
523 exchange estimated in Storfjorden to show that Arctic polynyas are likely a significant
524 sink for atmospheric CO₂. Our study confirms that at least one other Arctic polynya

525 behaves as they predict, an important step in validating their larger scale estimates. In
526 addition to polynyas, we hypothesize that flaw leads may act as important centres for
527 winter gas exchange. Leads are typically a small fraction of the Arctic icescape during
528 winter; *Lindsay and Rothrock* [1995] estimated the percentage to be 2–3% for the central
529 Arctic and 6–9% for the peripheral seas. However, our findings suggest that even at low
530 fractions these features may dominate the winter gas exchange budget much in the same
531 way that they dominate heat fluxes [*Maykut*, 1978; *Andreas*, 1980]. Also of note are the
532 large areas of the Arctic and Antarctic ocean which are seasonally ice-free. In the Arctic,
533 this makes up an area of 6.4×10^6 km² and in the Antarctic 15.2×10^6 km² (*Wadhams*
534 [2000], 1979-87 averages). As discussed by *Omar et al.* [2005], the seasonal formation of
535 sea ice over these areas may create short but intense CO₂ fluxes which could be important
536 to the annual air-sea CO₂ exchange budget of the Arctic and Southern Oceans.

537 Ongoing and anticipated changes in the polar oceans may further increase the impor-
538 tance of this effect. The rapidly decreasing summer ice extent in the Arctic (e.g. *Stroeve*
539 *et al.* [2007]) means that a larger area will be subject to annual ice formation, and signifi-
540 cant positive trends in sea ice motion [*Hakkinen et al.*, 2008] may create more winter-time
541 open water. Our results show that this will permit larger annual air-sea gas exchange, but
542 whether this will result in a larger net sink of CO₂ is complicated. Surface seawater that
543 is undersaturated in $p\text{CO}_{2sw}$ can only absorb a finite amount of CO₂, depending on the
544 state of the carbonate equilibrium (i.e. the Revelle factor). A debate is currently emerg-
545 ing regarding whether the ocean surface exposed by recent sea ice loss has the capacity
546 to take up significant amounts of CO₂ [*Bates*, 2006; *Cai et al.*, 2010]. A similar debate
547 needs to be had regarding uptake capacity of the Arctic Ocean at freeze-up in order to

548 understand the potential for gas exchange enhanced by ice formation. A net annual sink
549 also requires export of absorbed CO₂ to depth, a process that appears to occur effectively
550 on the shelves where deep water formation occurs but not necessarily in the Arctic Ocean
551 basins [Omar *et al.*, 2005]. Clearly, a lot of work remains to be done before we can fully
552 understand the interplay between enhanced gas exchange and future changes to the Arctic
553 Ocean.

6. Summary and Conclusions

554 This paper has provided the first direct, in situ observations of enhanced gas exchange
555 during sea ice formation. Eddy covariance calculations of CO₂ flux in Amundsen Gulf (a
556 polynya with a dynamic winter sea ice cover) showed periods of intense uptake (mean flux
557 -4.88 , maximum $-27.95 \mu\text{mol m}^{-2} \text{s}^{-1}$) and one case of outgassing (mean flux $+2.10 \mu\text{mol}$
558 $\text{m}^{-2} \text{s}^{-1}$). These periods of high gas exchange were observed coincidentally with high heat
559 fluxes, which we confirmed from satellite imagery to be the result of open water (i.e. flaw
560 leads). Conversely, we measured no fluxes above the uncertainty of our instruments over
561 consolidated sea ice.

562 We presented several hypotheses to explain our observations of enhanced gas transfer.
563 In a winter flaw lead, we expect high water-side turbulence to occur as a result of rapid
564 heat loss and salt rejection. Since turbulence is the first-order control on gas exchange,
565 we hypothesize that this high turbulence is a major cause of enhanced gas exchange. We
566 also discussed the modification of surface properties (temperature, salinity, DIC/TA) and
567 their effect on the seawater carbonate system. The potential of these modifications to
568 influence the rate of gas exchange depends on the saturation state of CO₂ with respect to
569 the atmosphere, and at times may actually be contradictory to high fluxes. In support of

570 these hypotheses, we were only able to provide limited evidence of high turbulence in the
571 region.

572 By comparing our flux values with DIC measurements we were able to show that al-
573 though high, they do fit within surface DIC budgets. A rough calculation of the integrated
574 CO₂ uptake over the months of Nov.–Jan. showed that winter gas exchange may in fact
575 be as important as the open water (i.e. late spring/summer/early fall) seasons. These
576 results have wide reaching implications for understanding the annual air–sea CO₂ budgets
577 of polynyas and other seasonally ice–free seas.

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References

- 594 Amiro, B. (2010). Estimating annual carbon dioxide eddy fluxes using open-path anal-
595 yzers for cold forest sites. *Agricultural and Forest Meteorology*, 150(10):1366–1372,
596 doi:10.1016/j.agrformet.2010.06.007.
- 597 Anctil, F., Donelan, M.A., Drennan, W.M., and Graber, H.C. (1994). Eddy-correlation
598 measurements of air-sea fluxes from a discus buoy. *Journal of Atmospheric and Oceanic*
599 *Technology*, 11:1144–1150.
- 600 Anderson, L.G., Falck, E., Jones, E.P., Jutterström, S., and Swift, J.H. (2004). Enhanced
601 uptake of atmospheric CO₂ during freezing of seawater: A field study in Storfjorden,
602 Svalbard. *Journal of Geophysical Research*, 109 (C06004), doi:10.1029/2003JC002120.
- 603 Andreas, E.L. (1980). Estimation of heat and mass fluxes over arctic leads. *Monthly*
604 *Weather Review*, 108:2957–2063.
- 605 Asher, W.E., Karle, L.M., Higgins, B.J., Farley, P.J., Monahan, E.C., and Leifer, I.S.
606 (1996). The influence of bubble plumes on air-seawater gas transfer velocities. *Journal*
607 *of Geophysical Research*, 101(C5):12,027-12,041, doi:10.1029/96JC00121.
- 608 Bates, N., Moran, S., Hansell, D., and Mathis, J. (2006). An increasing CO₂ sink in the
609 Arctic Ocean due to sea-ice loss. *Geophysical Research Letters*, 33:L23609
- 610 Bock, E.J., Hara, T., Frew, N.M., and McGillis, W.R. (1999). Relationship between air sea
611 gas transfer and short wind waves. *Journal of Geophysical Research*, 104(C11):25,821-
612 25,831, doi:10.1029/1999JC900200.

- 613 Bolin, B. (1960). On the exchange of carbon dioxide between the atmosphere and the sea.
614 *Tellus*, 12(3):274–281.
- 615 Bourgault, D., Kelley, D.E., and Galbraith, P.S. (2008). Turbulence and boluses on an
616 internal beach. *Journal of Marine Research*, 66:563–588.
- 617 Burba, G.G., McDermitt, D.K., Grelle, A.G., Anderson, D.J., and Xu, L. (2008). Ad-
618 dressing the influence of instrument surface heat exchange on the measurements of
619 CO₂ flux from open-path gas analyzers. *Global Change Biology*, 14(8):1854–1876, doi:
620 10.1111/j.1365-2486.2008.01606.x.
- 621 Cai, W., Chen, L., Chen, B., Gao, Z., Lee, S., Chen, J., Pierrot, D., Sullivan, K., Wang,
622 Y., Hu, X., et al. (2010). Decrease in the CO₂ uptake capacity in an ice-free Arctic
623 Ocean basin. *Science*, 329(5991):556.
- 624 Denman, K., Brasseur, G., Chidthaisong, A., Ciais, P., Cox, P., Dickinson, R., Hauglus-
625 taine, D., Heinze, C., Holland, E., Jacob, D., Lohmann, U., Ramachandran, S.,
626 da Silva Dias, P., Wofsy, S., and Zhang, X. (2007). Couplings between changes in the
627 climate system and biogeochemistry. In Solomon, S., Qin, D., Manning, M., Marquis,
628 M., Averyt, K., Tignor, M., and Miller, H., editors, *Climate change 2007: The Physical
629 Science Basis. Contribution of Working Group I to the Fourth Assessment Report of the
630 Intergovernmental Panel on Climate Change*, chapter 7. Cambridge University Press,
631 United Kingdom and New York, NY, USA.
- 632 Dugan, J.P., Panichas, S.L., and DiMarco, R.L. (1991). Decontamination of wind mea-
633 surements from buoys subjected to motions in a seaway. *Journal of Atmospheric and
634 Oceanic Technology*, 8:85–95.

- 635 Edson, J.B., Hinton, A.A., Prada, K.E., Hare, J.E., and Fairall, C.W. (1998). Direct
636 covariance flux estimates from mobile platforms at sea. *Journal of Atmospheric and*
637 *Oceanic Technology*, 15:547–562, doi: 10.1175/1520-0426(1998)015.
- 638 Frew, N.M. (1997). The role of organic films in air-sea gas exchange. In *The Sea Surface*
639 *and Global Change*, edited by P.S. Liss, R.A. Duce, *Cambridge University Press*, 121–
640 173.
- 641 Frew, N.M., Bock, E.J., Schimpf, U., Hara, T., Haußecker, H., Edson, J.B., McGillis,
642 W.R., Nelson, R.K., McKenna, S.P., Uz, B.M., Jähne, B. (2004). Air-sea gas transfer:
643 Its dependence on wind stress, small-scale roughness, and surface films. *Journal of*
644 *Geophysical Research*, 109(C8):C08S17, doi:10.1029/2003JC002131.
- 645 Fujitani, T. (1981). Direct measurement of turbulent fluxes over the sea during AMTEX.
646 *Papers in Meteorology and Geophysics*, 32(3):119–134.
- 647 Galley, R.J., Key, E., Barber, D.G., Hwang, B.J., and Ehn, J.K. (2008). Spatial and
648 temporal variability of sea ice in the southern Beaufort Sea and Amundsen Gulf: 1980–
649 2004. *Journal of Geophysical Research*, 113(C5):C05S95, doi:10.1029/2007JC004553.
- 650 Golden, K., Ackley, S., and Lytle, V. (1998). The percolation phase transition in sea ice.
651 *Science*, 282(5397):2238–2241.
- 652 Gosink, T.A., Pearson, J.G., and Kelley, J.J. (1976). Gas movement through sea ice.
653 *Nature*, 263:41–42, doi:10.1038/263041a0.
- 654 Hakkinen, S., Proshutinsky, A., and Ashik, I. (2008). Sea ice drift in the Arctic since the
655 1950s. *Geophysical Research Letters*, 35:L19704, doi:10.1029/2008GL034791.
- 656 Hirata, R., Hirano, T., Saigusa, N., Fujinuma, Y., Inukai, K., Kitamori, Y., Takahashi,
657 Y., and Yamamoto, S. (2007). Seasonal and interannual variations in carbon dioxide

- 658 exchange of a temperate larch forest. *Agricultural and Forest Meteorology*, 147(3-4):110–
659 124, doi:10.1016/j.agrformet.2007.07.005.
- 660 Ho, D.T., Law, C.S., Smith, M.J., Schlosser, P., Harvey, M., and Hill, P. (2006).
661 Measurements of air-sea gas exchange at high wind speeds in the Southern Ocean:
662 Implications for global parameterizations. *Geophysical Research Letters*, 33:L23503,
663 doi:10.1029/2006GL026817.
- 664 Ho, D.T., Zappa, C.J., McGillis, W.R., Bliven, L.F., Ward, B., Dacey, J.W., Schlosser, P.,
665 and Hendricks, M.B. (2004). Influence of rain on air-sea gas exchange: Lessons from a
666 model ocean. *Journal of Geophysical Research*, 109:C08S18, doi:10.1029/2003JC001806.
- 667 Jähne, B. (1987). On the parameters influencing air-water gas exchange. *Journal of*
668 *Geophysical Research*, 92(C2):1937–1950, 10.1029/JC092iC02p01937.
- 669 Järvi, L., Mammarella, I., Eugster, W., Ibrom, A., Siivola, E., Dellwik, E., Keronen,
670 P., Burba, G., and Vesala, T. (2009). Comparison of net CO₂ fluxes measured with
671 open-and closed-path infrared gas analyzers in an urban complex environment. *Boreal*
672 *Environment Research*, 14:499–514.
- 673 Kaimal, J.C. and Gaynor, J.E. (1991). Another look at sonic thermometry. *Boundary-*
674 *Layer Meteorology*, 56:401–410.
- 675 Kohsiek, W. (2000). Water vapor cross-sensitivity of open path H₂O/CO₂ sensors. *Journal*
676 *of Atmospheric and Oceanic Technology*, 17:299–311.
- 677 Körtzinger, A., Thomas, H., Schneider, B., Gronau, N., Mintrop, L., and Duinker,
678 J.C. (1996). At-sea intercomparison of two newly designed underway pCO₂
679 systems—encouraging results. *Marine Chemistry*, 52(2):133–145, doi:10.1016/0304-
680 4203(95)00083-6.

- 681 Kuss, J. and Schneider, B. (2004). Chemical enhancement of the CO₂ gas
682 exchange at a smooth seawater surface. *Marine Chemistry*, 91(1-4):165–174,
683 doi:10.1016/j.marchem.2004.06.007.
- 684 Leuning, R. (2004). Measurements of trace gas fluxes in the atmosphere using eddy
685 covariance: WPL corrections revisited. In *Handbook of micrometeorology: A guide to*
686 *surface flux measurement and analysis*, edited by Lee X., Massman W.J., Law B.E. ,
687 *Boston:Kluwer*, 119–32.
- 688 Lindsay, R.W. and Rothrock, D.A. (1995). Arctic sea ice leads from advanced very high
689 resolution radiometer images. *Journal of Geophysical Research*, 100(C3):4533–4544,
690 doi:10.1029/94JC02393.
- 691 Liss, P.S. and Merlivat, L. (1986). Air-sea gas exchange rates: Introduction and synthesis.
692 In *The role of air-sea exchange in geochemical cycling*, edited by Buat-Menard, P. ,
693 *Boston:D. Reidel Publishing Company*, 113-127
- 694 Loose, B., McGillis, W.R., Schlosser, P., Perovich, D., and Takahashi, T. (2009). Effects
695 of freezing, growth, and ice cover on gas transport processes in laboratory seawater
696 experiments. *Geophysical Research Letters*, 36(5):L05603, doi:10.1029/2008GL036318.
- 697 Loose, B., Schlosser, P., Perovich, D., Ringelberg, D., Ho, D., Takahashi, T., Richter-
698 Menge, J., Reynolds, C., McGillis, W., and Tison, J.-L. (2010). Gas diffusion through
699 columnar laborator sea ice: implications for mixed layer ventilation of co₂ in the seasonal
700 sea ice zone. *Tellus B*, pages doi: 10.1111/j.1600-0889.2010.00506.x.
- 701 Maykut, G.A. (1978). Energy exchange over young sea ice in the Central Arctic. *Journal*
702 *of Geophysical Research*, 83(C7):3646–3658, doi:10.1029/JC083iC07p03646.

- 703 McGillis, W.M., Edson, J.B., Ware, J.E., Dacey, J.H.W., Hare, J.E., Fairall, C.W., and
704 Wanninkhof, R. (2001). Carbon dioxide flux techniques performed during GasEx-98.
705 *Marine Chemistry*, 75:267–280, doi:10.1016/S0304-4203(01)00042-1.
- 706 McGillis, W.R., Edson, J.B., Zappa, C.J., Ware, J.D., McKenna, S.P., Terray, E.A., Hare,
707 J.E., Fairall, C.W., Drennan, W.M., Donelan, M., DeGrandpre, M.D., Wanninkhof, R.,
708 and Feely, R.A. (2004). Air-sea CO₂ exchange in the equatorial Pacific. *Journal of*
709 *Geophysical Research*, 109:C08S02, doi:10.1029/2003JC002256.
- 710 McPhee, M.G. and Stanton, T.P. (1996). Turbulence in the statically unstable oceanic
711 boundary layer under arctic leads. *Journal of Geophysical Research*, 101(C3):6409–6428.
- 712 Melling, H. and Moore, R. (1995). Modification of halocline source waters during freez-
713 ing on the Beaufort Sea shelf: evidence from oxygen isotopes and dissolved nutrients.
714 *Continental Shelf Research*, 15(1):89–113.
- 715 Mitsuta, Y. and Fujitani, T. (1974). Direct measurement of turbulent fluxes on a cruising
716 ship. *Boundary-Layer Meteorology*, 6:203–217, doi: 10.1007/BF00232485.
- 717 Mucci, A., Lansard, B., Miller, L.A., and Papakyriakou, T.N. (2010). CO₂ fluxes across
718 the air–sea interface in the southeastern Beaufort Sea: Ice-free period. *Journal of*
719 *Geophysical Research*, 115(C04003), doi:10.1029/2009JC005330.
- 720 Murata, A. and Takizawa, T. (2003). Summertime CO₂ sinks in shelf and slope
721 waters of the western Arctic Ocean. *Continental Shelf Research*, 23(8):753–776,
722 doi:10.1016/S0278-4343(03)00046-3.
- 723 Nightingale, P.D., Malin, G., Law, C.S., Watson, A.J., Liss, P.S., Liddicoat, M.I., Boutin,
724 J., and Upstill-Goddard, R.C. (2000). In situ evaluation of air-sea gas exchange param-
725 eterizations using novel conservative and volatile tracers. *Global Biogeochemical Cycles*,

- 726 14:373–387, doi:10.1029/1999GB900091.
- 727 Nomura, D., Yoshikawa-Inoue, H., and Toyota, T. (2006). The effect of sea-ice growth on
728 air–sea CO₂ flux in a tank experiment. *Tellus B*, 58(5):418–426.
- 729 Omar, A., Johannessen, T., Bellerger, R.G.J., Olsen, A., Anderson, L.G., and Kivimäe
730 (2005). Sea–ice and brine formation in Storfjorden: Implications for the Arctic win-
731 tertime air–sea CO₂ flux. In *The Nordic Seas: an Integrated Perspective*, edited by H.
732 Drange, T. Dokken, T. Furevik, R. Gerdes, W. Berger, *American Geophysical Union*,
733 *Geophysical Monograph 158*, 177–189.
- 734 Omstedt, A. (1985). On supercooling and ice formation in turbulent sea-water. *Journal*
735 *of Glaciology*, 31:263–271.
- 736 Ono, K., Miyata, A., and Yamada, T. (2008). Apparent downward CO₂ flux observed
737 with open-path eddy covariance over a non-vegetated surface. *Theoretical and Applied*
738 *Climatology*, 92(3):195–208, doi: 10.1007/s00704-007-0323-3.
- 739 Prytherch, J., Yelland, M.J., Pascal, R.W., Moat, B.I., Skjelvan, I., and Neill, C.C. (2010).
740 Direct measurements of the CO₂ flux over the ocean: Development of a novel method.
741 *Geophysical Research Letters*, 37(3):L03607, doi:10.1029/2009GL041482.
- 742 Rysgaard, S., Glud, R.N., Sejr, M.K., Bendtsen, J., and Christensen, P.B. (2007). Inor-
743 ganic carbon transport during sea ice growth and decay: A carbon pump in polar seas.
744 *Journal of Geophysical Research*, 112:C03016, doi:10.1029/2006JC003572.
- 745 Rysgaard, S., Bendtsen, J., Pedersen, L.T., Ramløv, H., Glud, R.N. (2009). Increased
746 CO₂ uptake due to sea ice growth and decay in the Nordic Seas. *Journal of Geophysical*
747 *Research*, 114:C09011, doi:10.1029/2008JC005088.

- 748 Shadwick, E., Thomas, H., Chierici, M., Else, B., Fransson, A., Michel, C., Miller, L.,
749 Mucci, A., Niemi, A., Papakyriakou, T., and J.-É., T. (2011). Seasonal variability of
750 the inorganic carbon system in the Amundsen Gulf region of the southeastern Beaufort
751 Sea. *Limnology and Oceanography*, 56(1), 303–322, doi:10.4319/lo.2011.56.1.0303.
- 752 Skogseth, R., Nilsen, F., and Smedsrud, L.H. (2009). Supercooled water in an Arctic
753 polynya: observations and modeling. *Journal of Glaciology*, 55(189):43–52.
- 754 Stroeve, J., Holland, M.M., Meier, W., Scambos, T., and Serreze, M. (2007). Arc-
755 tic sea ice decline: Faster than forecast. *Geophysical Research Letters*, 34(9):9501,
756 doi:10.1029/2007GL029703.
- 757 Sweeney, C., Gloor, E., Jacobson, A.R., Key, R.M., McKinley, G., Sarmiento, J.L.,
758 and Wanninkhof, R. (2007). Constraining global air-sea gas exchange for CO₂
759 with recent bomb ¹⁴C measurements. *Global Biogeochemical Cycles*, 7:843–878,
760 doi:10.1029/2006GB002784.
- 761 Takagaki, N. and Komori, S. (2007). Effects of rainfall on mass transfer
762 across the air-water interface. *Journal of Geophysical Research*, 112(C6):C06006,
763 doi:10.1029/2006JC003752.
- 764 Takahashi, T., Olafsson, J., Goddard, J.G., Chipman, D.W., and Sutherland, S.C. (1993).
765 Seasonal variation of CO₂ and nutrients in the high-latitude surface oceans: a compar-
766 ative study. *Global Biogeochemical Cycles*, 7(4):843–878, doi:10.1029/93GB02263.
- 767 Takahashi, T., Sutherland, S.C., Wanninkhof, R., Sweeney, C., Feely, R.A., Chipman,
768 D.W., Hales, B., Friederich, G., Chavez, F., Sabine, C., et al. (2009). Climatological
769 mean and decadal change in surface ocean pCO₂, and net sea-air CO₂ flux over the global
770 oceans. *Deep Sea Research Part II: Topical Studies in Oceanography*, 56(8-10):554–577,

- 771 doi:10.1016/j.dsr2.2008.12.009 .
- 772 Tsang, G. and Hanley, T. O'D. (1985). Frazil formation in water of different salinities
773 and supercoolings. *Journal of Glaciology*, 31(108).
- 774 Ushio, S. and Wakatsuchi, M. (1993). A laboratory study on supercooling and frazil
775 ice production processes in winter coastal polynyas. *Journal of Geophysical Research*,
776 98(C11):20321.
- 777 Wadhams, P. (2000). *Ice in the Ocean*. Gordon and Breach Science Publishers
- 778 Wanninkhof, R. (1992). Relationship between wind speed and gas exchange over the
779 ocean. *Journal of Geophysical Research*, 97(C5):7373–7382.
- 780 Wanninkhof, R., Asher, W.E., Ho, D.T., Sweeney, C., and McGillis, W.R. (2009). Ad-
781 vances in quantifying air-sea gas exchange and environmental forcing. *Annual Review*
782 *of Marine Science*, 1:213–244.
- 783 Wanninkhof, R. and McGillis, W.R. (1999). A cubic relationship between gas ex-
784 change and wind speed over the ocean. *Geophysical Research Letters*, 26:7373–7381,
785 doi:10.1029/1999GL900363.
- 786 Webb, E.K., Pearman, G.I., and Leuning, R. (1980). Correction of flux measurements for
787 density effects due to heat and water vapour transfer. *Quarterly Journal of the Royal*
788 *Meteorological Society*, 106(447):85–100, doi: 10.1002/qj.49710644707.
- 789 Woolf, D.K. (1997). Bubbles and their role in gas exchange. In *The Sea Surface and*
790 *Global Change*, edited by P.S. Liss, R.A. Duce, Cambridge University Press, 173–207.
- 791 Woolf, D.K. (2005). Parametrization of gas transfer velocities and sea-state-dependent
792 wave breaking. *Tellus B*, 57(2):87–94, doi: 10.1111/j.1600-0889.2005.00139.x.

- 793 Woolf, D.K., Leifer, I.S., Nightingale, P.D., Rhee, T.S., Bowyer, P., Caulliez, G.,
794 De Leeuw, G., Larsen, S.E., Liddicoat, M., Baker, J., and Andreae, M.O. (2007). Mod-
795 elling of bubble-mediated gas transfer: Fundamental principles and a laboratory test.
796 *Journal of Marine Systems*, 66(1-4):71–91, doi:10.1016/j.jmarsys.2006.02.011.
- 797 Yager, P.L., Wallace, D.W.R., Johnson, K.M., Smith Jr., W.O., Minnett, P.J., Dem-
798 ington, J.W. (1995). The Northeast Water Polynya as an atmospheric CO₂ sink: A sea-
799 sonal rectification hypothesis. *Journal of Geophysical Research*, 100(C3):4389–4398,
800 doi:10.1029/94JC01962.
- 801 Zappa, C.J., Asher, W.E., Jessup, A.T., Klinke, J., and Long, S.R. (2004). Microbreaking
802 and the enhancement of air-water transfer velocity. *Journal of Geophysical Research*,
803 109(C08S16):21, doi:10.1029/2003JC001897.
- 804 Zappa, C.J., Ho, D.T., McGillis, W.R., Banner, M.L., Dacey, J.W.H., Bliven, L.F., Ma,
805 B., and Nystuen, J. (2009). Rain-induced turbulence and air-sea gas transfer. *Journal*
806 *of Geophysical Research*, 114(C07009), doi:10.1029/2008JC005008.
- 807 Zappa, C.J., McGillis, W.R., Raymond, P.A., Edson, J.B., Hintsa, E.J., Zemmeling, H.J.,
808 Dacey, J.W.H., and Ho, D.T. (2007). Environmental turbulent mixing controls on
809 air-water gas exchange in marine and aquatic systems. *Geophysical Research Letters*,
810 34(10):10601, doi:10.1029/2006GL028790.

Table 1. Summary of conditions experienced during each sample case

Table 2. Noise and bias in the eddy covariance system, including the effect of various corrections. Bias is calculated as the mean CO₂ flux from cases where near-zero flux is expected, noise is one standard deviation around that mean. The number of eddy covariance sample runs is 274 for raw and sensor heating corrected, 151 for crosstalk corrected.

Table 3. Summary of monthly lead fraction, CO₂ fluxes and resulting change in mixed layer DIC

Figure 1. Map of the Banks Island flaw lead/polynya complex. The light grey line shows the ship track. The shaded grey area represents the region which usually remains mobile through the time period under consideration, and the dotted line shows the areas typically associated with the Cape Bathurst polynya and flaw lead.

Figure 2. Measured CO₂ fluxes (including sensor heating correction) for the study period. The numbers along the top axis indicate the sample cases, the most interesting of which are discussed in the text, with the red brackets denoting their time frame. The inset shows observations made between Dec. 1–2 with an extended scale on the y-axis. The horizontal grey lines show the estimated noise level of the eddy covariance system as discussed in section 5.1

Figure 3. RADARSAT-1 image collected on Nov. 2 at 01:53, just prior to sample case 1. The inset map shows the location of the imaged area, red lines indicate the ship’s track, the green X indicates the location of the ship at the time of image acquisition, and the green arrow shows the mean wind direction.

Figure 4. Time series of atmospheric measurements made during sample case 1. (a) Measured CO₂ flux with sensor heating correction added (open circles), and the estimated noise level of the system as discussed in section 5.1 (horizontal grey lines), (b) measured sensible heat flux (red open circles) and latent heat flux (blue open circles), (c) 1 minute averages of air temperature (dashed line) and wind velocity (solid line).

Figure 5. RADARSAT-1 image collected on Nov. 20 at 01:29, just prior to cases 3 and 4. The inset map shows the location of the imaged area, red lines indicate the ship’s track, the green X indicates the location of the ship at the time of image acquisition, and the green arrow shows the mean wind direction.

Figure 6. Time series of atmospheric measurements made during sample cases 3 and 4 (division between the two cases is denoted by the dashed vertical line). (a) Measured CO₂ flux with sensor heating correction added (open circles), range of bulk CO₂ flux estimates (brackets), and the estimated detection limit of the system as discussed in section 4.2 (horizontal grey lines), (b) measured sensible heat flux (red open circles) and latent heat flux (blue open circles), (c) 1 minute averages of air temperature (dashed line) and wind velocity (solid line).

Figure 7. RADARSAT-1 image collected on Dec. 1, 14:45, just after case 7. The inset map shows the location of the imaged area, red lines indicate the ship’s track, the green X indicates the location of the ship at the time of image acquisition, and the green arrow shows the mean wind direction.

Figure 8. Time series of atmospheric measurements made during sample case 7. (a) Measured CO₂ flux with sensor heating correction added (open circles), and the estimated noise level of the system as discussed in section 5.1 (horizontal grey lines), (b) measured sensible heat flux (red open circles) and latent heat flux (blue open circles), (c) 1 minute averages of air temperature (dashed line) and wind velocity (solid line).

Figure 9. RADARSAT-1 image collected on Jan. 24, 01:33, just prior to case 17. The inset map shows the location of the imaged area, red lines indicate the ship’s track, the green X indicates the location of the ship at the time of image acquisition, and the green arrow shows the mean wind direction.

Figure 10. Time series of atmospheric measurements made during sample case 17. (a) Measured CO₂ flux with sensor heating correction added (open circles), and the estimated noise level of the system as discussed in section 5.1 (horizontal grey lines), (b) measured sensible heat flux (red open circles) and latent heat flux (blue open circles), (c) 1 minute averages of air temperature (dashed line) and wind velocity (solid line).

Figure 11. Time series of atmospheric measurements made during sample case 10. (a) Measured CO₂ flux with sensor heating correction added (open circles), and the estimated noise level of the system as discussed in section 5.1 (horizontal grey lines), (b) measured sensible heat flux (red open circles) and latent heat flux (blue open circles), (c) 1 minute averages of air temperature (dashed line) and wind velocity (solid line).

Figure 12. Schematic summarizing the important processes occurring during a wind-driven lead event. The processes highlighted in blue/red are those which likely have a direct effect on air–sea gas exchange. Processes in red are associated with frazil ice formation, and those in blue are associated with the surface cooling.

Figure 13. Open water percentage for Amundsen Gulf (122–126°W, 70–71.5°N) during the study period, as determined by classification of near-weekly RADARSAT-1 imagery.

Case	Date	Location (Lat/Lon)	CO ₂ Flux ($\mu\text{mol m}^{-2} \text{s}^{-1}$)	ΔCO_2 (μatm)	H Flux (W m^{-2})	E Flux (W m^{-2})	Air T ($^{\circ}\text{C}$)	Wind Vel. (m)	Wind Dir. (deg)	Sea Ice Conditions
1	11/02 04:30- 11/03 09:30	71.185 ^a -129.096	-1.81	-80	43.3	4.1	-7.5	8.7	29	Newly forming grease ice
2	11/08 02:15- 11/09 00:50	69.498 -123.930	0.23	146.5	7.3	3.7	-18.9	5.5	131.1	Newly forming fast ice. Estimated thickness: 30-40cm
3	11/20 01:30- 11/20 14:45	71.038 -123.297	2.1	-77.4	53.8	9.9	-15.2	12. 5	98.4	Mobile ice with upwind leads, thickness: 37cm
4	11/20 16:00- 11/20 18:30	71.071 -123.430	-9.58	-66.7	111.4	11.3	-14.2	11. 8	110.8	Mobile ice with upwind leads, thickness: 37cm
5	11/28 07:30- 11/29 02:00	70.419 ^a -126.372	0.55	15.8	-2.3	0.8	-16	8	254.6	Consolidated mobile ice, thickness: 52cm
6	11/30 05:15- 11/30 23:30	71.053 ^a -123.954	-0.03	-52.1	1.3	-1.1	-15.8	11. 5	314.1	Consolidated, ridged ice floe

7	12/01 07:00- 12/01 12:30	71.590 ^a -124.656	-26.88	N/A	33.4	-10.4	-16.6	7.3	3.4	Transit through active lead with open water and grease ice
8	12/01 13:45- 12/02 02:45	71.901 -125.441	0.31	-63.6	-2.9	0	-19.7	5.1	37.3	Land fast ice
9	12/02 05:30- 12/02 22:15	71.725 -125.597	0.35	-69.4	1.3	-0.3	-18.2	3.4	47.1	Consolidated ice floe, thickness: 35cm
10	12/04 21:00- 12/06 12:15	71.402 ^a -124.875	-0.09	-86.7	15.8	2	-18	5.1	267.3	Consolidated ice floes, varying thicknesses: 25– 45cm
11	12/19 23:15- 12/22 18:15	71.915 -125.433	0.42	-49.8	-0.6	3.3	-22	4.6	108.8	Land fast ice
12	12/24 20:45- 12/25 17:15	71.262 -124.383	-0.14	-51	5.5	1.3	-20.7	5.9	123.2	Consolidated mobile ice, thickness: 30cm
13	01/02 18:15- 01/06 03:30	71.306 ^a -124.722	-0.5	-52	-7.9	5.3	-21.8	11. 6	117.9	Thick consolidated ice floe, thickness: 105cm

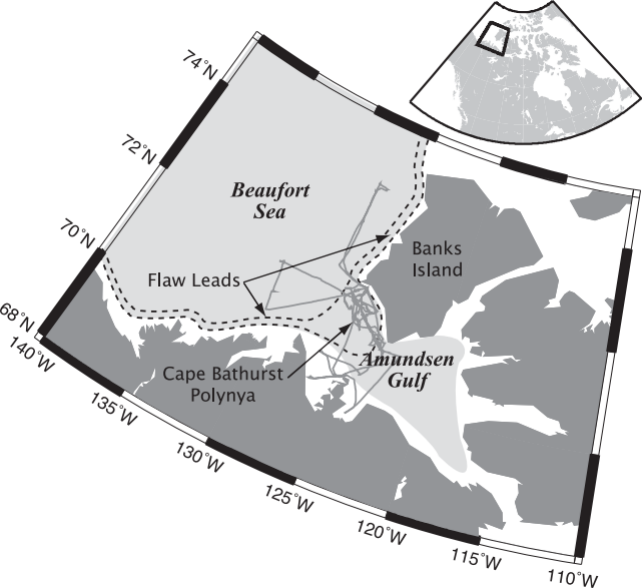
14	01/10 09:15- 01/11 18:45	71.653 ^a -126.101	0.58	-78.3	2.9	1.2	-21.3	7	123.3	Consolidated mobile ice
15	01/13 16:00- 01/14 22:00	71.494 ^a -124.638	0.34	-73.8	3.2	1	-25	8	62.4	Consolidated mobile ice
16	01/20 17:00- 01/21 04:30	71.579 -125.104	0.34	N/A	-2.9	2.8	-18.9	6.4	123.6	Consolidated mobile ice, thickness: 91cm
17	01/24 08:00- 01/25 05:30	71.203 -125.184	-3.15	-38.4	16.4	5.6	-20.3	11. 9	291.7	Consolidated mobile ice, upwind leads
18	01/25 18:00- 01/26 16:30	71.172 -125.014	-0.86	-8.6	9.4	2	-25.7	8.7	298.5	Consolidated mobile ice
	^a Ship was in transit, or drifting significantly: reported value is the midpoint of the sampling period									

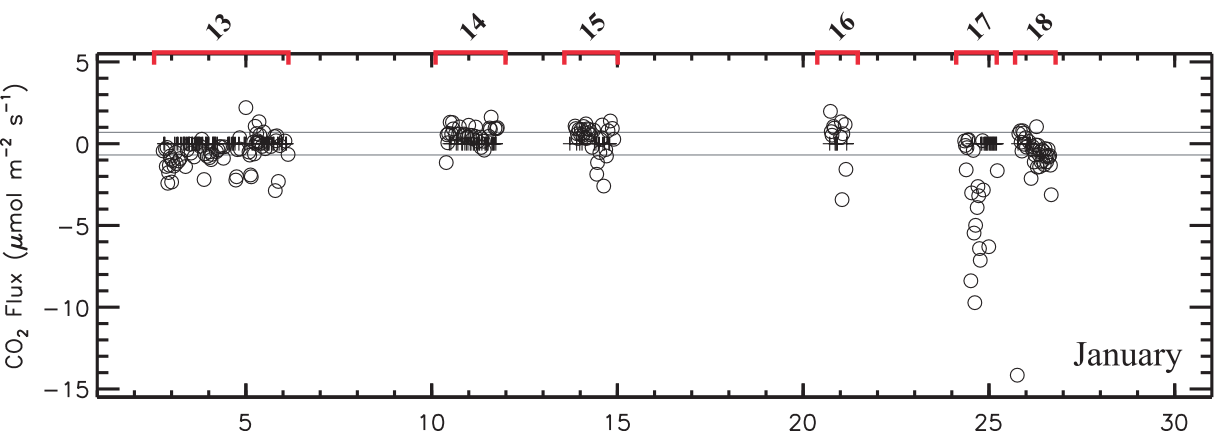
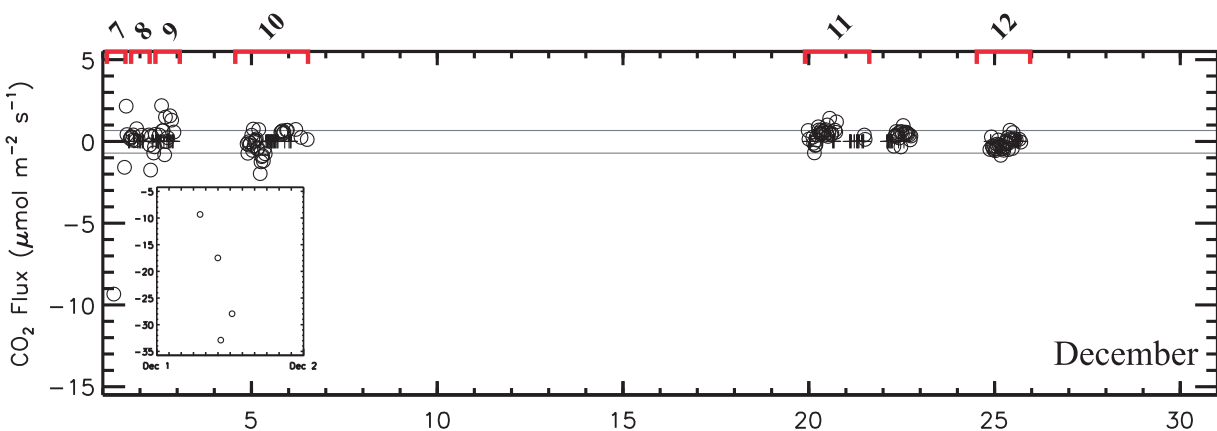
	Bias ($\mu\text{mol m}^{-2} \text{s}^{-1}$)	Noise ($\mu\text{mol m}^{-2} \text{s}^{-1}$)
Raw (uncorrected)	-0.45	± 0.76
Sensor heating corrected	0.13	± 0.77
Water vapour crosstalk correction	-0.21	± 1.32

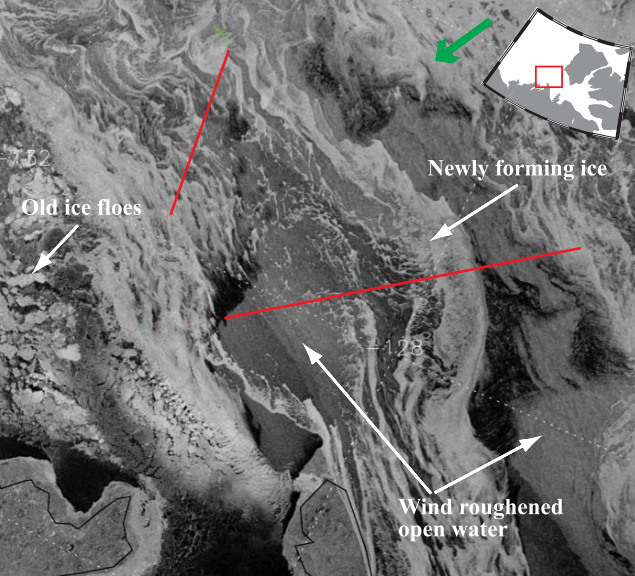
	November	December	January
^a Mean Open Water (%)	6.4	4.1	1.2
^b FCO ₂ _{sw-mon} (μmol m ⁻² s ⁻¹)	-0.3	-0.2	-0.1
^c ΔDIC _{as-enh} (μmol kg ⁻¹)	16.2	10.7	3.1
^d ΔDIC _{as} (μmol kg ⁻¹) [Shadwick et al., 2011]	1	2	0.1
^e ΔDIC _{bio} (μmol kg ⁻¹) [Shadwick et al., 2011]	3.0 ± 10.0	13.0 ± 10.0	16.0 ± 10.0

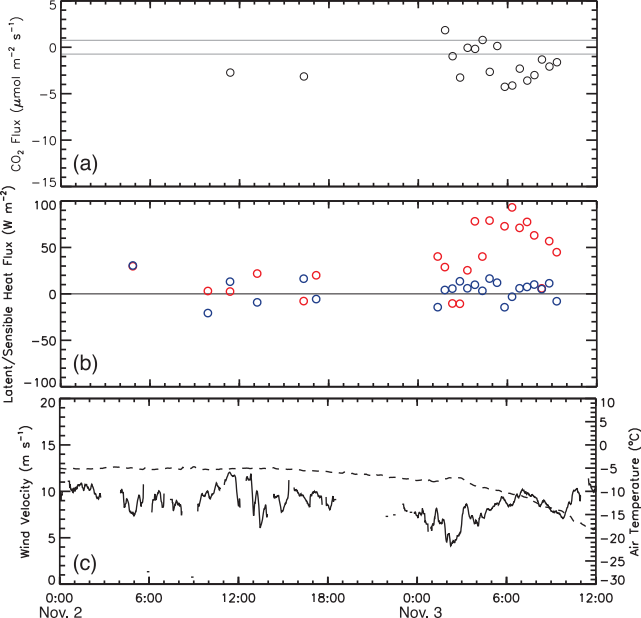
^a Calculated from RADARSAT-1 image classification, ^b calculated from mean of cases 1,4,7 & 17, multiplied by lead fraction and integrated over the month, ^c calculated change in DIC concentration over a 50m mixed-layer using FCO₂_{sw-mon},

^d calculated change in DIC concentration over a 50 m mixed-layer using bulk-flux estimates scaled for open water fraction (from Shadwick et al. [2011]), ^e calculated change in DIC concentration over a 50m mixed layer due to biological activity (from Shadwick et al. [2011]).







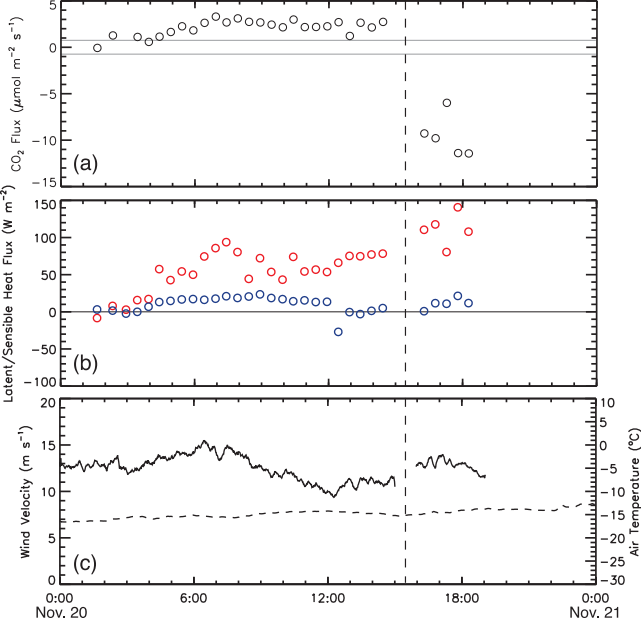


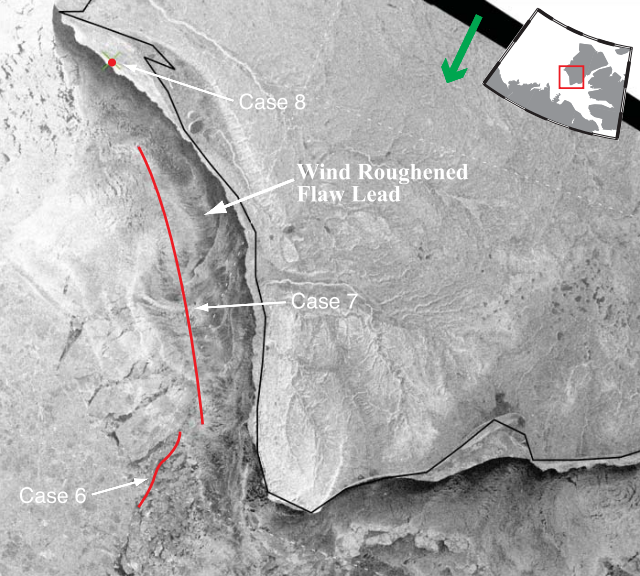


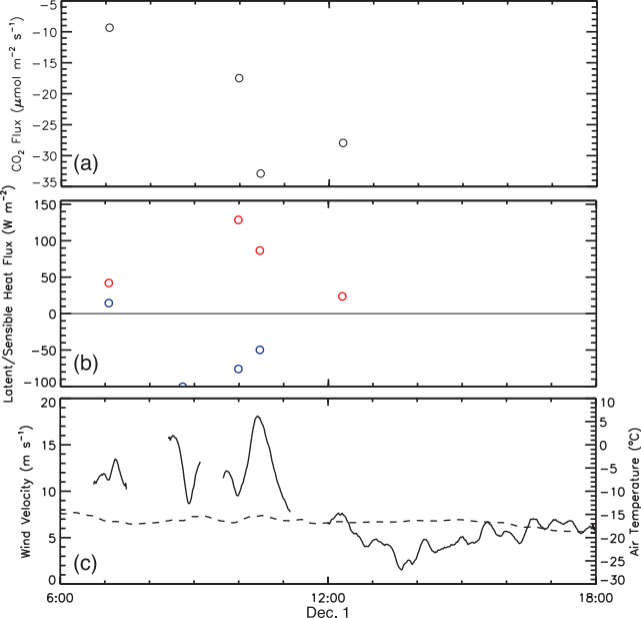
Flaw leads

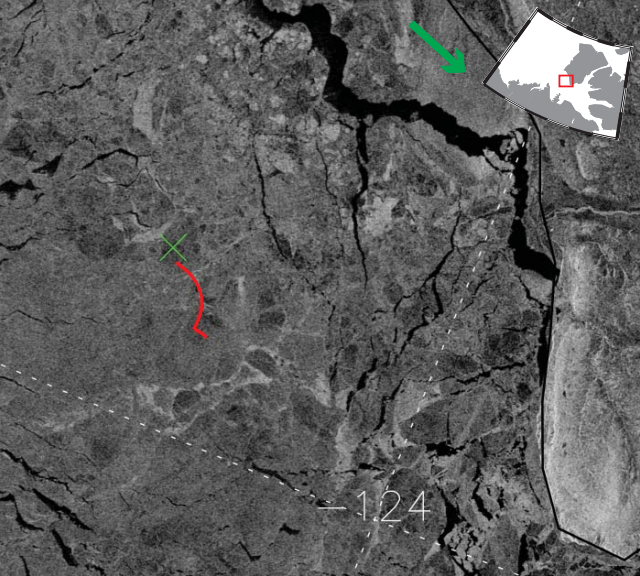
Case 4

Case 3









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