

# 1 Evaluation of the National Oceanic and Atmospheric Administration/ 2 Coupled-Ocean Atmospheric Response Experiment (NOAA/COARE) 3 air-sea gas transfer parameterization using GasEx data

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7 [1] During the two recent GasEx field experiments, direct covariance measurements of  
8 air-sea carbon dioxide fluxes were obtained over the open ocean. Concurrently, the  
9 National Oceanic and Atmospheric Administration/Coupled-Ocean Atmospheric  
10 Response Experiment air-sea gas transfer parameterization was developed to predict gas  
11 transfer velocities from measurements of the bulk state of the sea surface and atmosphere.  
12 The model output is combined with measurements of the mean air and sea surface carbon  
13 dioxide fugacities to provide estimates of the air-sea CO<sub>2</sub> flux, and the model is then tuned  
14 to the GasEx-1998 data set. Because of differences in the local environment and possibly  
15 because of weaknesses in the model, some discrepancies are observed between the  
16 predicted fluxes from the GasEx-1998 and GasEx-2001 cases. To provide an estimate of  
17 the contribution to the air-sea flux of gas due to wave-breaking processes, the whitecap  
18 and bubble parameterizations are removed from the model output. These results show that  
19 moderate (approximately 15 m s<sup>-1</sup>) wind speed breaking wave gas transfer processes  
20 account for a fourfold increase in the flux over the modeled interfacial processes. *INDEX*  
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## 30 1. Introduction

31 [2] The air-sea exchange of climate relevant compounds  
32 and, in particular, carbon dioxide has come under increased  
33 scrutiny because of the continued uncertainty surrounding  
34 the mass sequestration of climate relevant compounds into  
35 the world's oceans and because of the potential consequence  
36 of atmospheric increases of these compounds on the Earth's  
37 climate. Mesoscale and larger-scale models could potentially  
38 supply reasonable regional- to global-scale estimations of  
39 the total gas transfer, provided that an accurate small-scale  
40 parameterization of the interfacial gas flux is incorporated.

41 Focused surface process studies of gas transfer over the open  
42 ocean hold promise to provide information that can be used  
43 to improve the gas transfer parameterizations. Although  
44 techniques are improving, comprehensive open ocean mea-  
45 surement of the air-sea carbon dioxide flux and of the  
46 processes relevant to gas transfer present significant diffi-  
47 culties for measurements made from a ship.

48 [3] An ideal parameterization would accurately incorpo-  
49 rate all of the physical mechanisms of gas transfer, includ-  
50 ing surface processes (influence of waves, microscale and  
51 larger-scale wave breaking, interfacial thermal structure,  
52 etc), subsurface processes (bubbles and turbulent mixing),  
53 micrometeorological influences (atmospheric surface layer  
54 turbulent structure, wind gustiness, etc), and would also  
55 account for chemical and marine biological sources and  
56 sinks. Much of the physics of gas exchange and the relative  
57 importance of each physical process remain largely un-  
58 known. However, it is imperative that progress in the  
59 development of accurate gas transfer parameterizations  
60 and in the continued improvement of measurement technol-  
61 ogy proceed in parallel. Of particular importance to gas  
62 exchange is the relative contribution of the globally aver-  
63 aged flux that occurs at relatively higher wind speeds. 64  
64 Because of wave-breaking and bubble mediation processes,

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65 gas transfer may be significantly enhanced in these regimes,  
66 yet direct field observations are nearly nonexistent.

67 [4] Before the advent of direct measurement of air-sea  
68 gas fluxes, the simple bulk relationship most commonly  
69 used in numerical models was of the form:

$$F = V_t \Delta X, \quad (1)$$

71 where  $F$  represents the flux,  $V_t$  is the gas transfer velocity  
72 (typically derived empirically), and  $\Delta X$  is the sea-air  
73 concentration difference. The first estimates of the transfer  
74 velocity were obtained from wind-water tank studies [e.g.,  
75 *Kanwisher*, 1963; *Liss and Merlivat*, 1986] and global  
76 isotopic distributions [*Broecker and Peng*, 1974]. Although  
77 these pioneering studies advanced our understanding of the  
78 processes controlling air-water gas exchange, there persists  
79 a need to develop a parameterization that captures all of the  
80 relevant physical processes in open ocean gas transfer and  
81 also accurately represents the timescale of the forcing  
82 mechanisms.

83 [5] Research has firmly established that the air-sea gas  
84 transfer velocity (and therefore the gas flux) is a function of  
85 wind speed [*Broecker et al.*, 1986; *Jähne et al.*, 1987], but  
86 the numerous conflicting empirical relationships have  
87 triggered controversy over the years [*Wanninkhof*, 1992;  
88 *Smith and Jones*, 1985; *Liss and Merlivat*, 1986]. This  
89 problem has persisted, although recent developments have  
90 begun to reconcile our understanding of the relationship  
91 between wind speed and gas transfer [*McGillis et al.*,  
92 2001a]. However, because of the difficulties of making  
93 measurements in the open ocean [*Fairall et al.*, 2000], there  
94 is a dearth of data at moderate to high wind speeds, where it  
95 is expected that a significant portion of the gas exchange  
96 takes place [*Wanninkhof and McGillis*, 1999; *Spillane et al.*,  
97 1986]. For example, during GasEx-1998, winds up to  
98  $17 \text{ m s}^{-1}$  were recorded, but only for brief periods [*McGillis*  
99 *et al.*, 2001a].

100 [6] Despite the paucity of direct gas flux observations in  
101 moderate to high wind speeds, there is a considerable  
102 theoretical basis for parameterizing gas flux in this regime.  
103 Air-entraining surface wave breaking is initiated at about  
104  $6\text{--}7 \text{ m s}^{-1}$  [*O'Muircheartaigh and Monahan*, 1986]. This  
105 process affects the surface roughness and introduces bub-  
106 bles into the ocean mixed layer, thereby locally bypassing  
107 the direct interfacial gas exchange. These effects may  
108 dramatically increase the air-sea gas transfer. An additional  
109 contribution arises from enhanced turbulence beneath the  
110 breakers. For example, *Terray et al.* [1996] observed  
111 significant enhancement of turbulent dissipation in the  
112 presence of breaking waves. This effect is also expected  
113 to increase the gas flux, but a lack of near-surface ocean  
114 observations impedes our understanding of the physical  
115 processes. Some insights into gas transfer in the presence  
116 of breaking waves have been gleaned from laboratory  
117 studies [*Asher and Wanninkhof*, 1998], but there are signif-  
118 icant scaling problems that prevent direct translation of the  
119 empirical results into geophysical scales.

120 [7] Because of the uncertainty surrounding gas transfer  
121 physics and the lack of observations, climate models em-  
122 ploy a wide variety of transfer velocity parameterizations,  
123 which leads to significant ambiguity for estimates of the  
124 globally integrated carbon dioxide sequestration. For exam-

125 ple, using a Weibul global wind distribution, a simple  
126 comparison of the *Wanninkhof and McGillis* [1999] cubic  
127 relationship to the *Wanninkhof* [1992] quadratic relationship  
128 yields a near doubling of the globally integrated annual flux  
129 of  $\text{CO}_2$  into the ocean. This difference will have severe  
130 consequences in long-term climate model estimates of the  
131 thermal influence of climate relevant compounds in the  
132 atmosphere.

133 [8] In section 2, we will briefly describe the GasEx set of  
134 experiments, and the measurements that were used in the  
135 evaluation herein. Section 3 contains a brief description of  
136 the National Oceanic and Atmospheric Administration/  
137 Coupled-Ocean Atmospheric Response Experiment  
138 (NOAA/COARE) gas flux parameterization, while more  
139 attention is given to the specifics of the computation of  
140 the gas flux in section 4. In section 5, we present the results  
141 of the application of the parameterization to the GasEx data  
142 sets, and we compare the output of the parameterized gas  
143 fluxes to the direct covariance flux measurements. Finally, a  
144 brief conclusion is provided in section 6.

## 2. GasEx Experiments 145

146 [9] In 1998, the Ocean-Atmosphere Carbon Exchange  
147 Study (OACES) program (now part of the Global Carbon  
148 Cycle program) of the NOAA Office of Global Programs  
149 (OGP) initiated a program of process studies intended to  
150 improve our understanding of air-sea gas flux processes.  
151 These investigations were designed to make observations of  
152 gas fluxes and gas transfer forcing mechanisms on relatively  
153 short timescales (1 hour) with the goal to quantify gas  
154 transfer velocities through improved parameterizations.

155 [10] The first deployment occurred onboard the NOAA  
156 ship *Ronald H. Brown* in May and June 1998. This multi-  
157 institutional, interdisciplinary air-sea experiment was named  
158 GasEx-1998. The primary study occurred in the North  
159 Atlantic in a warm-core eddy near  $46^\circ\text{N}$ ,  $21^\circ\text{W}$ , and this  
160 locale was specifically selected to provide a stable labora-  
161 tory for injection of deliberate tracers and for maximization  
162 of  $\text{CO}_2$  transfer signal levels. Current gas instrumentation  
163 technology limits our ability to detect carbon dioxide  
164 fluctuations associated with transfer processes except in  
165 the largest source or sink regimes. The stability of the eddy  
166 and the presence of an algal bloom led to significant  
167 atmospheric signal levels of carbon dioxide over the course  
168 of the main experiment, with a mean value of the air-sea  
169 partial pressure gradient,  $\Delta p\text{CO}_2$ , of approximately  
170  $-85 \mu\text{atm}$ . More details are given by *McGillis et al.*  
171 [2001a, 2001b].

172 [11] During GasEx-1998, a modified fast response, closed  
173 path, nondispersive infrared (NDIR)  $\text{CO}_2/\text{H}_2\text{O}$  gas analyzer  
174 was deployed on the ship to continuously measure atmo-  
175 spheric carbon dioxide and water vapor fluctuations. When  
176 combined with motion-corrected sonic anemometer mea-  
177 surements of local fluctuations of the vertical wind velocity,  
178  $w'$  [*Edson et al.*, 1998], direct covariance estimates of the  
179 air-sea flux of carbon dioxide were obtained as in:

$$F = \overline{w'c'}, \quad (2)$$

181 where the overbar denotes a time average, and  $w'$  and  $c'$   
182 are the fluctuations of vertical velocity and gas concen-



183 tration, respectively. Additional instruments deployed on  
184 the ship were used to make measurements of the local  
185 mean meteorological and surface conditions: air tempera-  
186 ture and humidity, sea surface temperature, downwelling  
187 solar and infrared radiative flux, cloud base height, and  
188 atmospheric boundary layer profiles of temperature,  
189 humidity, and wind. Continuous samples of atmospheric  
190 and oceanic concentrations of CO<sub>2</sub> were made with the  
191 permanent NOAA Atlantic Oceanic and Meteorological  
192 Laboratory (AOML) system on the ship.

193 [12] On the basis of the success of the first GasEx cruise  
194 [McGillis *et al.*, 2001a, 2001b], another surface processes  
195 and gas flux campaign was scheduled for February 2001.  
196 This cruise also took place on the NOAA ship *Ronald H.*  
197 *Brown* and was named GasEx-2001. In contrast to the first  
198 GasEx-1998 cruise, this expedition was primarily located  
199 in the eastern Pacific, just south of the equatorial Pacific  
200 upwelling region. Complex processes and iron deficiency  
201 limit the biological productivity in this region [Strutton *et*  
202 *al.*, 2004], so this region is a relatively strong source of  
203 CO<sub>2</sub> into the atmosphere. The average ΔpCO<sub>2</sub> for GasEx-  
204 2001 was approximately +110 μatm, with a characteristic  
205 diurnal cycling of about 5 μatm. On the basis of measure-  
206 ments of the local microlayer, little biological activity was  
207 detected at the sea surface [Nelson Frew, WHOI personal  
208 communication].

209 [13] As in the first GasEx cruise, an improved closed path  
210 NDIR system was deployed to measure the fluctuations of  
211 CO<sub>2</sub> in order to compute the direct covariance flux. The  
212 mean meteorological and CO<sub>2</sub> measurements were essen-  
213 tially the same for the second deployment, with the addition  
214 of measurements of the underlying waves and atmospheric  
215 boundary layer wind profiles from the permanent 915 MHz  
216 radar system [Law *et al.*, 2002]. The wind profiler and  
217 ceilometer measurements confirmed that the mesoscale  
218 meteorological conditions during the course of the experi-  
219 ment were very steady, with very little variability in  
220 boundary layer structure.

### 221 3. NOAA/COARE Air-Sea Gas Flux 222 Parameterization

223 [14] Recently, Fairall *et al.* [2000] presented a gas  
224 transfer parameterization, which is based on the well-  
225 known COARE Bulk Flux Algorithm [Fairall *et al.*,  
226 1996b] with the addition of surface renewal concepts  
227 from Soloviev and Schluessel [1994]. The original COARE  
228 model contains an algorithm for the oceanic cool skin,  
229 which has been generalized for gas transfer applications.  
230 A full presentation of the development of the NOAA/  
231 COARE gas flux parameterization, in addition to relevant  
232 background information on the history of gas transfer  
233 parameterization and micrometeorological measurement  
234 techniques are given by Fairall *et al.* [2000]. Ideally, a  
235 physically based air-sea gas transfer parameterization will  
236 require reasonably available variables as inputs. That is,  
237 the algorithm should only require input variables that can  
238 be readily measured over the relevant driving scales from  
239 in situ or remote sensors. In addition, the parameterization  
240 should compute the gas transfer velocity within the  
241 context and timescale of the relevant environmental  
242 variables and processes (radiative and turbulent fluxes,

wind speed, wave state, surface current, air-sea tempera- 243  
ture difference, near-surface water thermal structure, etc). 244  
Fairall *et al.* [2000] present one such micrometeorologi- 245  
cally based air-sea gas transfer parameterization. 246

[15] The parameterization relies on matching of the water 247  
and air flux expressions (both of which are expressed in 248  
terms of molecular and turbulent components), in addition 249  
to attending to the details of the molecular layer transfer on 250  
the water side. The final expression is quite general, and can 251  
be applied to any gas: 252

$$F_s = \frac{A_{sol} u_{*a} \Delta p_x}{\sqrt{\rho w / \rho a} [h w S_{cw}^{1/2} + \ln(z w / \delta w) / \kappa] + \alpha [h a S_{ca}^{1/2} + C_d^{1/2} - 5 + \ln(S_{ca}) / (2\kappa)]}, \quad (3)$$

where the subscripts *s*, *a*, and *w* denote the surface, air, and 254  
water, α is the dimensionless solubility (a function of 255  
species, temperature, and salinity), *u*<sub>\**a*</sub> is the air-side friction 256  
velocity, Δ*p*<sub>*x*</sub> is the partial pressure difference of the gas 257  
(species subscript *x*) across the air-sea interface, ρ is the 258  
density, *S*<sub>*c*</sub> is the Schmidt number of the gas, *z* is the depth 259  
of the measurement, δ is the estimated turbulent surface 260  
layer thickness, κ is the von Kármán constant, and *C*<sub>*d*</sub> is the 261  
atmospheric velocity drag coefficient. In addition, in 262  
equation (3) the solubility has been expressed as 263

$$A_{sol} = 10^5 \alpha / (R_{gas} T), \quad (4)$$

where *T* is the temperature, *R*<sub>*gas*</sub> is the universal gas 265  
constant, and 266

$$h = \frac{\Lambda R_r^{1/4}}{\varphi}, \quad (5)$$

where Λ is an adjustable constant, *R*<sub>*r*</sub> is the roughness 268  
Reynolds number, and φ is an empirical function that 269  
accounts for buoyancy effects on turbulent transfer in the 270  
ocean. Most of the computed variables in equation (3) are 271  
estimated from the NOAA/COARE algorithm, given the 272  
input of air and sea temperature, wind speed, specific 273  
humidity, salinity, downwelling shortwave and longwave 274  
radiation, rain rate, atmospheric pressure, and measure- 275  
ment heights and water depth. Details on the computation 276  
of these variables are given by Fairall *et al.* [1996b, 277  
2000]. In fact, the practical calculation from equation (3) 278  
can be isolated to estimate the solubility times gas 279  
transfer velocity (α*k*<sub>*co2*</sub>) by dividing both sides of the 280  
equation by the partial pressure difference. This informa- 281  
tion (Δ*p*<sub>*x*</sub>) can be obtained from the current generation of 282  
continuous underway air and sea CO<sub>2</sub> measurement 283  
systems on ships [Wanninkhof and Thoning, 1993]. Thus 284  
the NOAA/COARE gas flux parameterization provides a 285  
description of the physical environs, and we require an 286  
external estimate of the gas mass differential across the 287  
interface in order to arrive at the gas flux. 288

[16] A number of possible sea ‘surface’ temperatures 289  
inputs exist: subsurface (3–5 m) water intake temperature 290  
from the ship thermosalinograph, the near-surface temper- 291  
ature measured from a floating thermistor (or modeled 292  
from the 5 m temperature), or a radiometrically measured 293  
skin temperature [Ward *et al.*, 2004]. The molecular 294

295 conditions near the interface, including the interfacial skin  
 296 temperature, control the mass flux. The cool skin has a  
 297 temperature that is primarily controlled by evaporation  
 298 and net longwave radiative cooling, and its depth is  
 299 confined to the molecular diffusive sublayer in the water.  
 300 Below this depth, there may occur a warm layer whose  
 301 structure is determined by the balance of solar warming  
 302 of the upper ocean and mixing processes. Deep, strong  
 303 gradient warm layers tend to occur in light wind con-  
 304 ditions. If measurements of the true interfacial tempera-  
 305 ture are not available, then the use of any other measured  
 306 temperature as input to the bulk algorithm requires a  
 307 model of the thermal structure up to the skin surface  
 308 [Fairall *et al.*, 1996b].

309 [17] The NOAA/COARE air-sea bulk gas transfer param-  
 310 eterization makes use of turbulence scaling theory applied  
 311 to both fluids, and matches the fluid models at the interface  
 312 where the suppression of the smallest turbulent eddies is  
 313 accomplished via viscous dissipation. Bubble mediation and  
 314 wave breaking have only been addressed in the model in a  
 315 heuristic fashion because of the complex nature of these  
 316 physical forcings. Given the lack of scientific consensus on  
 317 this issue, we have simply chosen an additive bubble and  
 318 wave-breaking enhancement for the gas transfer velocity  
 319 [Woolf, 1997]:

$$k_b = Vf\alpha^{-1} \left[ 1 + (e\alpha S_c^{-1/2})^{-1/n} \right]^{-n}, \quad (6)$$

321 where  $V$ ,  $e$ , and  $n$  are constants, and the whitecap fraction,  $f$ ,  
 322 from Monahan and O’Muircheartaigh [1980], is given by

$$f = BU^{3.4}, \quad (7)$$

323 where  $U$  is the mean wind speed and  $B$  is an empirical  
 324 constant. This higher wind enhancement of the transfer  
 325 velocity in equation (6) is added to the transfer velocity,  
 326  $k_{co2}$ , derived from equation (3) to establish the combined  
 327 effect of interfacial and breaking wave processes.  
 328

#### 329 4. Gas Flux

330 [18] The expression in equation (3) of the air-sea flux of  
 331 carbon dioxide (or any gas) can be written as

$$F = k\alpha(fCO_{2w} - fCO_{2a}), \quad (8)$$

333 where  $k$  is the gas transfer velocity (similar to  $V_i$  in equation  
 334 (1)), and  $fCO_2$  is the fugacity of carbon dioxide in the bulk  
 335 water and ambient air, respectively. Practically, measure-  
 336 ment of fugacity at the air-sea interface is not possible, so  
 337 water is drawn in situ from the ‘bulk surface’ water at  
 338 depths between 1–5 m. The solubility is a function of both  
 339 temperature and salinity and can be empirically described  
 340 such as by Wanninkhof [1992, Table A2]. The solubility for  
 341 CO<sub>2</sub> varies by a few percent per degree around 20°C [Weiss,  
 342 1974]. The gas transfer velocity can be conceptualized as  
 343 related to the traditional transfer coefficients in bulk  
 344 turbulent flux parameterizations. This quantity has recently  
 345 been expressed as a quadratic and cubic function of wind  
 346 speed ( $U$ ) in Wanninkhof [1992] and Wanninkhof and

McGillis [1999], respectively, and it also has a Schmidt 347  
 number dependence: 348

$$k = k(Sc^{-n}, U^m), \quad (9)$$

where  $n$  varies from 0.67 for a smooth surface to 0.4 for a 350  
 surface characterized with bubbles, and  $m$  has been 351  
 empirically determined to be between 2 and 3 for field data. 352

[19] In most ship-based situations, the water-side CO<sub>2</sub> 353  
 mixing ratio is measured by drawing continuous bulk 354  
 (subscript  $w$ ) water samples into a headspace equilibrator 355  
 [Wanninkhof and Thoning, 1993]. Typically, this sample 356  
 comes from a water intake in the forward hull of the ship at 357  
 a depth of 3–5 m. The air-side sample (from a height of 358  
 approximately 10 m) and headspace carbon dioxide mixing 359  
 ratios are determined with an NDIR detector, and the results 360  
 are converted to fugacity. Given direct measurement of the 361  
 gas flux [McGillis *et al.*, 2001a] coincident with air and sea 362  
 gas mixing ratio measurements, one can use equation (8) to 363  
 determine the gas transfer velocity. This quantity has been 364  
 used to develop simplified wind speed-dependent gas 365  
 exchange models [Wanninkhof and McGillis, 1999]. 366

[20] Upon closer inspection, the flux is more accurately 367  
 expressed in terms of the bulk and interfacial solubilities as 368  
 in 369

$$F = k_s(\alpha_w fCO_{2w} - \alpha_s fCO_{2a}), \quad (10)$$

where the subscript  $s$  indicates that the transfer velocity and 371  
 solubility are computed at the skin temperature and salinity. 372  
 The expression in equation (10) assumes that mass is 373  
 conserved (i.e.,  $\alpha_w fCO_{2w}$  is conserved with depth), which 374  
 is a simplification assuming no biological effects [Ward *et* 375  
*al.*, 2004]. In fact, this expression ignores the warm layer 376  
 effect in the ocean [Fairall *et al.*, 1996a], while we require 377  
 that the water-side fugacity and solubility be computed at 378  
 the surface, since the interfacial characteristics so strongly 379  
 affect the gas flux. For example, cool skin temperature 380  
 depression typically ranges from 0.1 to 0.4 degrees [Fairall 381  
*et al.*, 1996a], and warm layer elevation under light winds 382  
 can be as large as 2°–3° in the tropics. This thermal 383  
 structure from the bulk water up to the surface will certainly 384  
 alter the flux. For the sake of physical accuracy in the 385  
 parameterization, we will express the flux in terms of the 386  
 interfacial characteristics. 387

[21] There are other thermal influences on the flux of the 388  
 gas, including the effect of the warm layer (and/or cool skin) 389  
 on the fugacity of CO<sub>2</sub> dissolved in seawater [Takahashi *et* 390  
*al.*, 1993] via the carbonate reaction. This requires a 391  
 deviation from the mass conservation assumption, so the 392  
 surface fugacity is expressed in terms of the bulk water 393  
 fugacity through an empirical relationship: 394

$$fCO_{2s} = fCO_{2w}(1 + 0.0423\Delta T), \quad (11)$$

where  $\Delta T = T_w - T_s$ . The small percentage change with 395  
 temperature was determined at a reference temperature of 397  
 20°C, and it also includes the effect of solubility changes 398  
 with temperature. A quick computation of the solubility at 399  
 20°C reveals that the effect of temperature alone on 400  
 solubility accounts for approximately 2.7% of the total 401  
 change in fugacity. Therefore the remaining 1.5% change in 402



403 fugacity of carbon dioxide due to a temperature difference  
 404 between the bulk water and the interfacial surface comes  
 405 from the change in mass due to reaction. We have chosen to  
 406 maintain the structure of our expression of the flux as in  
 407 equation (10) with the addition of the small temperature  
 408 correction to the mass due to the carbonate reaction as  
 409 follows:

$$F = k_s \alpha_s \left[ fCO_{2w} \frac{\alpha_w}{\alpha_s} (1 + 0.015\Delta T) - fCO_{2a} \right], \quad (12)$$

411 Here,  $\alpha_w$  and  $\alpha_s$  are evaluated at temperatures  $T_w$  and  $T_s$ ,  
 412 respectively.

413 [22] This expression is more accurate than equation (8)  
 414 and provides a statement of the flux in terms of the  
 415 measured fugacities and bulk water temperature, along with  
 416 the estimate of skin temperature. The NOAA/COARE  
 417 algorithm models the skin temperature by combining both  
 418 the cool skin and warm layer algorithms applied to measure-  
 419 ments within or below the warm layer from which the  
 420 surface solubility can be computed. For the GasEx data sets,  
 421 the error represented by calculation of the flux using  
 422 equation (10) instead of the more accurate equation (12)  
 423 is at most a few percent [Hare *et al.*, 2003]. However,  
 424 under circumstances of strong cool skin and warm layer  
 425 development, the necessity of computing the flux from  
 426 equation (12) becomes more imperative.

427 [23] From Soloviev and Schlüssel [1994], the water-side  
 428 component of equation (5) is modified to the expression

$$hw = \frac{\Lambda R_r^{1/4}}{A \varphi_x}, \quad (13)$$

430 where  $\varphi_x$  is a function providing for buoyancy effects  
 431 [Fairall *et al.*, 2000],  $\Lambda$  is the cool skin adjustment constant  
 432 equal to 13.3, and  $A$  has a value of 1.85 [Soloviev and  
 433 Schlüssel, 1994] based on a supersaturated radon data set.  
 434 The NOAA/COARE parameterization also allows for the  
 435 substitution of the COARE cool-skin algorithm to char-  
 436 acterize the surface, but the physics are essentially the same.  
 437 In the COARE model, the cool skin constant has been  
 438 empirically found to be 15.8. The difference between the  
 439 Soloviev and Schlüssel [1994] and Fairall *et al.* [2000]  
 440 constants will play a part in the gas transfer parameteriza-  
 441 tion of the two GasEx data sets in section 5.

442 [24] Alternatively, the algorithm allows for input of direct  
 443 measurement of the radiometric (skin) temperature, if it is  
 444 available. In addition, the algorithm models the depth of the  
 445 warm layer, which can be used to determine whether  
 446 the fugacity measurement has been made within or below  
 447 the layer. It is necessary to add a thermal correction to the  
 448 measurements in the conversion from mole fraction to  
 449 fugacity [Ward *et al.*, 2004], but we will not include those  
 450 details here.

451 [25] The current version of the NOAA/COARE gas flux  
 452 parameterization also provides an estimate of the surface  
 453 gas transfer velocity expressed as the sum of the influential  
 454 physical resistances, which include the molecular and  
 455 turbulent components on both sides of the interface. An  
 456 additional transfer velocity term arises under higher wind  
 457 regimes as a result of bubble mediation and wave breaking  
 458 and is expressed in equation (6). No considerations are  
 459 given to the effect of biological processes or to surfactants,

460 although these environmental forcings are recognized to be  
 461 potentially significant [Frew, 1997].

## 5. Application of the Parameterization 462

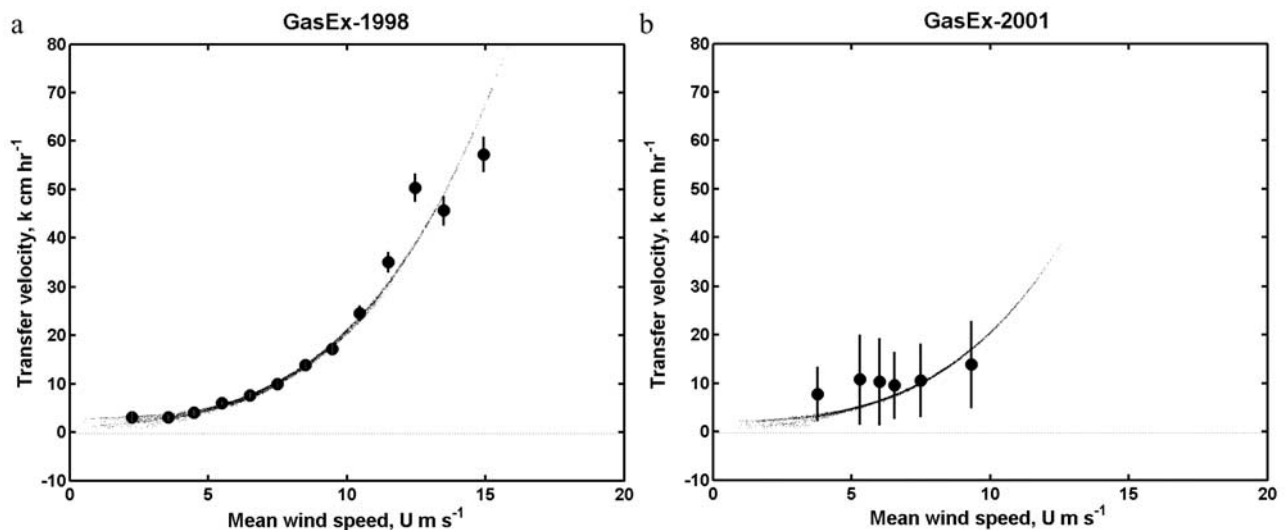
[26] The NOAA/COARE gas transfer parameterization 463  
 464 has been evaluated using data collected during the GasEx-  
 465 1998 and GasEx-2001 experiments (see section 2). These  
 466 data include directly measured carbon dioxide fluxes, mean  
 467 meteorological and surface observations, and water and air  
 468 CO<sub>2</sub> fugacities. Standard procedures were used to exclude  
 469 data from undesirable relative wind directions, during ship  
 470 maneuvers or rain, for occasional poor motion corrections  
 471 or abnormal values in the observations, etc. We then input  
 472 the resulting 10 min averaged data into the NOAA/COARE  
 473 routine, and after tuning the model to the GasEx-1998 data  
 474 set, we are able to evaluate the resulting balance between  
 475 interfacial and breaking wave processes at higher winds. In  
 476 addition, a comparison of the parameterized transfer veloci-  
 477 ties from the two GasEx data sets will shed some light into  
 478 the model's weaknesses.

[27] We know of no other gas transfer model which 479  
 480 incorporates the level of detailed physics which is included  
 481 in the NOAA/COARE parameterization. Nevertheless, we  
 482 realize that not all of the complex interfacial processes are  
 483 adequately represented in the model. For example, the  
 484 bubble contribution to the flux relies on a simple wind  
 485 speed-dependent empirical formula, and there are no pro-  
 486 visions in the model for the effects of biological processes.  
 487 In the following sections, we will examine some of the  
 488 deficiencies of the NOAA/COARE parameterization in  
 489 order to highlight areas for future improvement. Then, we  
 490 will demonstrate the effect that wave-breaking and bubble  
 491 processes have on the gas flux through an analysis of the  
 492 interfacial component of the parameterization and the di-  
 493 rectly measured GasEx-1998 fluxes. Finally, we will exam-  
 494 ine the overall error in the measured and parameterized  
 495 results by compositing the GasEx results in terms of the  
 496 interfacial fugacity discontinuity. This analysis also includes  
 497 data from very low flux regimes obtained during the 1999  
 498 season in the western Pacific.

### 5.1. GasEx Parameterizations 499

[28] Initially,  $A$  in equation (13) and  $V$  in equation (6) 500  
 501 were tuned to the GasEx-1998 flux results. We chose to  
 502 adjust these coefficients to the GasEx-1998 data set because  
 503 of the wide range of wind speeds observed during this  
 504 cruise compared to that of the GasEx-2001 expedition.  
 505 The fit to the data requires adjustment of the two coeffi-  
 506 cients ( $A = 0.625$  and  $V = 4900$ ) which are significantly  
 507 different than those provided from the original references  
 508 (Soloviev and Schlüssel [1994],  $A = 1.85$ ; and Woolf [1997],  
 509  $V = 2450$ ).

[29] Given the circumstances, the modification of the high 510  
 511 wind speed coefficient is not particularly alarming. In the  
 512 case of the Woolf [1997] model, the coefficient,  $V$ , was  
 513 derived from an expression which was best fit to modeled  
 514 bubble transfer velocities and was combined with statistical  
 515 estimates of a wind speed-dependent model of whitecap  
 516 coverage from photographic evidence [O'Muircheartaigh  
 517 and Monahan, 1986]. Neither of these empirical expressions  
 518 can be considered to be comprehensive, given the wide



**Figure 1.** Modeled (small points) and measured (solid circles, from equation (12)) CO<sub>2</sub> transfer velocity versus mean wind speed for (a) GasEx-1998 and (b) GasEx-2001. The NOAA/COARE gas transfer parameterization has been tuned for best fit to this data set.

519 range of wind, wave, and breaking conditions which could  
 520 be encountered in the field. Furthermore, numerous simpli-  
 521 fying assumptions have been made in the development of the  
 522 *Woolf* [1997] and *O’Muircheartaigh and Monahan* [1986]  
 523 expressions, which contribute to the considerable uncertain-  
 524 ties which surround them.

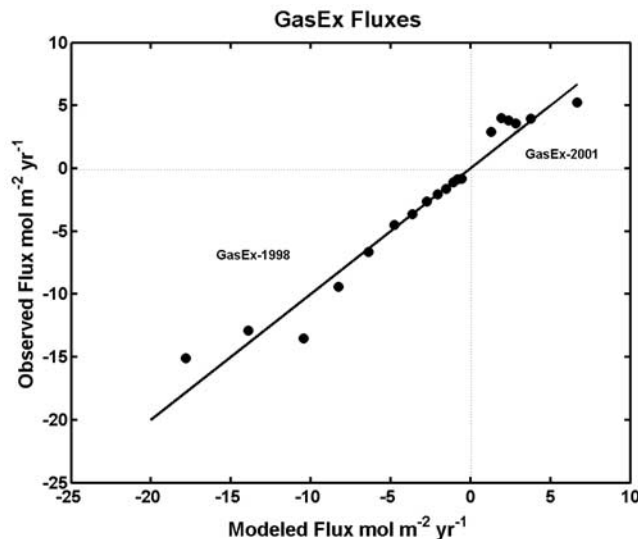
525 [30] Moreover, the original low wind speed empirical  
 526 constant,  $A$ , [*Soloviev and Schlüssel*, 1994, Figure 4] was  
 527 obtained from a combination of data sets which displayed  
 528 an admitted large degree of uncertainty. The radon field data  
 529 used to infer their coefficient was obtained indirectly and  
 530 had been integrated over several days of sampling with all  
 531 environmental variability embedded within. Given that our  
 532 parameterization was applied to 10 min mean samples, it is  
 533 difficult to interpret the resulting adjustment to  $A$ . Finally, it  
 534 must be recognized that GasEx-1998 was conducted in a  
 535 particular biological and physical field environment, which  
 536 may or may not be considered to be representative of a  
 537 typical gas transfer regime.

538 [31] Figure 1 shows the transfer velocity plotted versus  
 539 mean wind speed for GasEx-1998 (Figure 1a) and GasEx-  
 540 2001 (Figure 1b). The small points are the optimized  
 541 NOAA/COARE gas flux parameterization output from  
 542 equation (3) including the enhanced transfer velocity  
 543 contribution from equation (6), and the large solid circles  
 544 are the bin-averaged transfer velocities computed from  
 545 the actual CO<sub>2</sub> flux and mean measurements and using  
 546 equation (12). The small points demonstrate more scatter  
 547 at low wind speeds, which indicates more sensitivity to  
 548 buoyant processes than for those points at higher winds.  
 549 From Figure 1a, reasonable agreement is seen between the  
 550 model-derived transfer velocity and the actual measurements.  
 551 This is not unexpected, since we have tuned the two model  
 552 parameters for best agreement to the 1998 data set. However,  
 553 an examination of the 2001 results (Figure 1b) shows some  
 554 significant deviation between the NOAA/COARE param-  
 555 eterization output and the measurements, particularly for  
 556 lower ( $<6$  m s<sup>-1</sup>) winds. The range of wind speeds was

557 limited during GasEx-2001, with the majority of the mea-  
 558 sured wind speeds between 4 and 7 m s<sup>-1</sup>. Fewer samples are  
 559 available for winds above the threshold for wave breaking,  
 560 so we cannot speculate on the significance of the higher  
 561 wind speed bin model fit for the GasEx-2001 results.

562 [32] From Figure 1, we see relatively greater gas transfer  
 563 at low wind speeds near the Equator (GasEx-2001). Some  
 564 significant differences were observed between the two  
 565 experimental locales, which might explain the inconsistent  
 566 fit to the model. Significant biological activity was observed  
 567 during GasEx-1998 (based on the measured flux of CO<sub>2</sub>  
 568 into the sea surface), while GasEx-2001 was in a regime of  
 569 low biological productivity [*Strutton et al.*, 2004]. In  
 570 addition, GasEx-2001 took place in the Equatorial current,  
 571 where shear mechanics and diurnal cycling force variability  
 572 of the ocean mixed layer depth between 2 and 20 m  
 573 (W. McGillis et al., Air-sea CO<sub>2</sub> exchange in the equatorial  
 574 Pacific, submitted to *Journal of Geophysical Research*,  
 575 2004). In contrast, GasEx-1998 occurred in a midlatitude  
 576 stable warm-core eddy with little diurnal variability. Signif-  
 577 icant mixing occurred as a result of changes in wind speed  
 578 throughout the course of the GasEx-1998 experiment  
 579 [*McGillis et al.*, 2001a]. The NOAA/COARE param-  
 580 eterization characterizes the gas transfer from the mixed layer  
 581 through standard Monin-Obukhov similarity (MOS), which  
 582 works well on the air side but is oversimplified for the  
 583 ocean. This lack of detail may contribute to discrepancy  
 584 between the 1998 and 2001 results.

585 [33] A subsequent analysis reveals that a best fit of the  
 586 parameterization to the relatively low wind observed flux  
 587 data of GasEx-2001 requires the constant  $A$  to have a value  
 588 of approximately 1.5 (not shown), which is much closer to  
 589 the value surmised by *Soloviev and Schlüssel* [1994]. It  
 590 must be noted that the error bars on the GasEx-2001 transfer  
 591 velocity estimates are larger, and this can be seen from a  
 592 comparison of Figures 1a and 1b. The fact that the coeffi-  
 593 cient,  $A$ , requires readjustment to fit the GasEx-2001 data  
 594 highlights the significant uncertainty still inherent in the



**Figure 2.** Modeled (solid line) and directly measured (solid circles)  $\text{CO}_2$  fluxes from both GasEx experiments plotted versus the modeled flux.

parameterization of the transfer velocity and underlines the need for more comprehensive experimental examination of the factors which influence gas transfer at all wind speeds.

[34] For example, the NOAA/COARE algorithm assumes that all of the wind stress goes into production of mechanical turbulence in the water-side mixed layer. That is, the expression:

$$\rho_a u_{*a}^2 = \rho_w u_{*w}^2. \quad (14)$$

This is a simplification, as a portion of the wind energy is recognized to be transferred into the production and growth of waves as well as into driving the surface current. Furthermore, there is a reasonable expectation that the pressure field is modified in the presence of the surface wave field [Janssen, 1999] and that breaking waves are responsible for enhanced subsurface turbulent energy dissipation [Terray et al., 1996]. Both of these processes modify the balance of terms in the turbulent kinetic energy equation, and this modification of the near surface physics may create a situation in which the departure from MOS is a source of error in the model. However, these departures would be most influential in the moderate and high wind speed regimes.

[35] The fine-scale thermal structure of the water column is complicated by biological activity, cool skin and warm layer dynamics, and currents. The thermal profile from the depth of gas measurement at 5 m up to the surface will affect the resulting  $\text{CO}_2$  fugacity at the interface and will contribute to the buoyant production of turbulent energy in the water. All of these effects contributed to the oceanic thermal structure to various degrees during the GasEx cruises, but few comparable direct measurements were obtained for both experiments. Given that the thermal structure and biological activity are most influential to the gas flux in lighter wind situations, we surmise that these processes are the cause of the bias seen between Figures 1a

and 1b. The carbon dioxide flux during GasEx-1998 can be attributed to the biological productivity, although no direct productivity measurements were made during the cruise. In the case of Figure 1b, the lack of biological activity during GasEx-2001 contributes in the correct manner to the bias observed for the lower wind speeds.

[36] The surfactant levels present during GasEx-1998 and GasEx-2001 are likely to be the source of some of the disparity in the gas transfer values [Bock et al., 1999]. Microbial production in surface ocean water is a known source of colored dissolved organic matter (CDOM). Microbial production in the North Atlantic is higher than in the equatorial Pacific, and the CDOM levels are three times higher in the North Atlantic [Siegel et al., 2002]. This increase in CDOM and the corresponding increase in surface films could cause a factor of two decrease in gas exchange [Frew, 1997], which is consistent with the differences seen in Figures 1a and 1b.

[37] Figure 2 shows bin-averaged fluxes from both the NOAA/COARE parameterization (using equation (3)) and the directly measured covariance fluxes [McGillis et al., 2001a]. The solid 1:1 line is the modeled flux, and the large solid circles are the WHOI data. The ability of this model to represent the fluxes, at least in an averaged sense, is impressive for the GasEx-1998 data set. The GasEx-2001 parameterized results show somewhat more scatter, but the overall fit is reasonable.

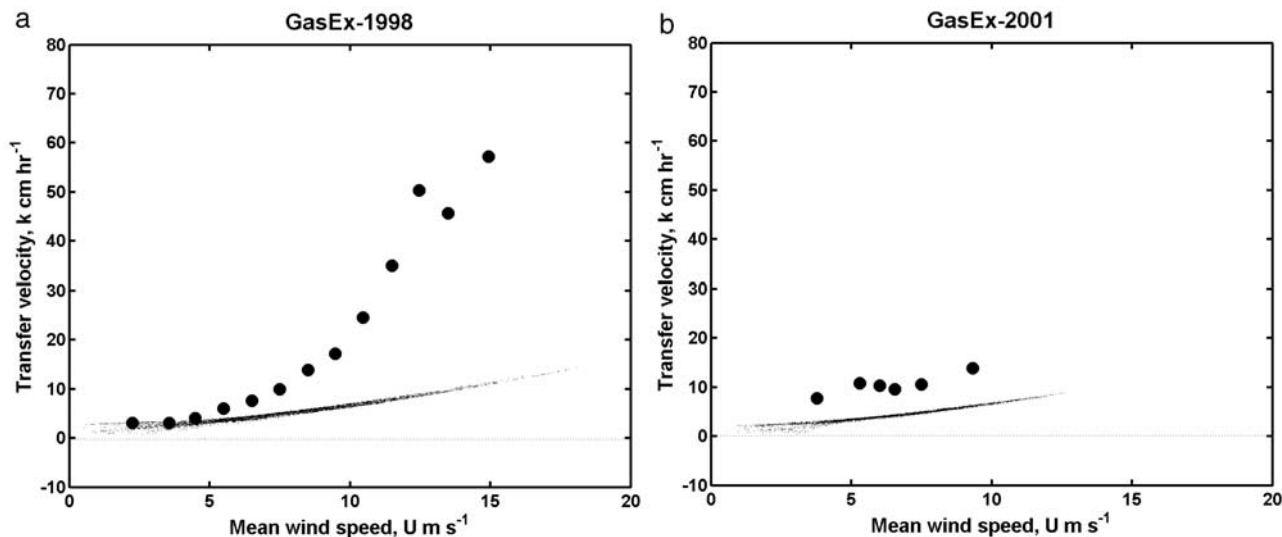
## 5.2. Whitecap Contribution

[38] Figure 3 shows transfer velocity plotted against mean wind speed for both experiments, but with the bubble parameterization omitted ( $k_b = 0$ ). This demonstration is enlightening, as it clearly shows the regime where bubbles and whitecap processes begin to affect the flux ( $U > 6 \text{ m s}^{-1}$ ) and shows the relative contribution to the transfer velocity at higher wind speeds. Here, the contrast between GasEx-1998 and GasEx-2001 becomes more striking. The equatorial region is apparently characterized by much stronger direct interfacial transfer but smaller bubble mediated effects, relative to the GasEx-1998 results.

[39] Further enlightenment is gained by examination of the fluxes in Figure 4, where the wave-breaking and bubble parameterization has been set to zero. Significant differences are seen between the parameterized fluxes (solid 1:1 line) and the measured fluxes (large solid circles) except in the lightest wind cases (below about  $2 \text{ mol m}^{-2} \text{ yr}^{-1}$ ). The lack of relatively high wind speed gas fluxes in the GasEx-2001 data set precludes making general statements about the universality of the breaking wave and bubble mediation enhancement of the fluxes. However, from the GasEx-1998 fluxes, we see that there is a factor of four enrichment of the  $\text{CO}_2$  transfer at approximately  $15 \text{ m s}^{-1}$ . This represents a significant finding for the comparison of the interfacial wave-breaking processes represented by both the NOAA/COARE gas flux parameterization and the directly measured fluxes from GasEx-1998. In addition, this observation highlights the need to improve air-sea gas flux parameterizations for the higher wind speeds.

[40] In Figure 5, we show the contribution to the air-sea gas flux from breaking wave processes alone. This estimate is obtained from the directly measured gas flux minus the interfacial flux, obtained from the parameterized transfer





**Figure 3.** Modeled (small points) and measured (solid circles) CO<sub>2</sub> transfer velocity versus mean wind speed for (a) GasEx-1998 and (b) GasEx-2001 with  $k_b = 0$  in equation (6).

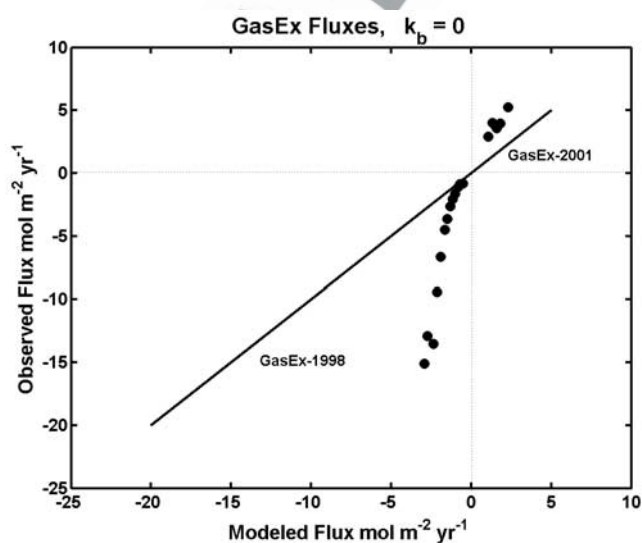
691 velocity with the whitecap/bubble algorithm turned off  
 692 ( $k_b = 0$  in equation (6)). Here, we plainly see the impact  
 693 of wave breaking to gas transfer, with more than 80% of the  
 694 flux at 14 m s<sup>-1</sup> coming from the breaking wave processes.

### 695 5.3. Three Cruise Composite

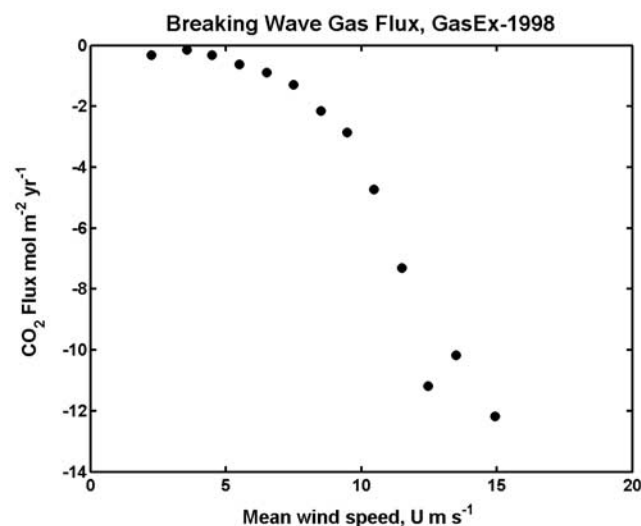
696 [41] In an effort to make improvements to the gas  
 697 instrumentation and measurement methods and to gain  
 698 experience with the system in a variety of environments,  
 699 NOAA/ETL and WHOI collaborated to deploy the NDIR  
 700 on the JASMINE and Nauru'99 cruises on the *Ronald H.*  
 701 *Brown* from May to July of 1999. The ship track for these  
 702 cruises ran from Singapore into the Indian Ocean, then  
 703 down to Darwin Australia, through the Solomon Islands,  
 704 and on to the island nation of Nauru at 0.32°S, 166.55°E.

The gas flux measurement effort was made as an ancillary 705  
 addition to the JASMINE and Nauru'99 objectives, and 706  
 more information for these two cruise legs can be found at 707  
 websites: <http://paos.colorado.edu/~jasmine/> and [http://](http://www.etl.noaa.gov/programs/1999/nauru99/) 708  
[www.etl.noaa.gov/programs/1999/nauru99/](http://www.etl.noaa.gov/programs/1999/nauru99/). 709

[42] For the first time, the CO<sub>2</sub> gas flux measurement 710  
 system was operated in a very low flux regime, as the mean 711  
 air-sea fugacity difference over the course of these cruises 712  
 was approximately 10 μatm. This low signal presents 713  
 significant challenges to the measurement of the carbon 714  
 dioxide flux, and careful scrutiny of the resulting data set 715  
 was required in order to obtain meaningful statistics. For 716  
 example, the presence of any rainfall, traveling close to 717  
 nearby islands, or modest ship motion produced unaccept- 718  
 able errors in the results. Using experience from GasEx- 719

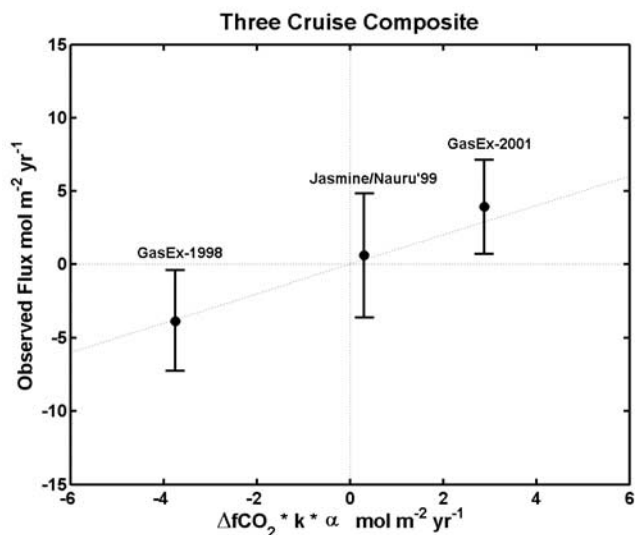


**Figure 4.** Modeled (solid line) and measured (solid circles) CO<sub>2</sub> fluxes from both GasEx experiments with  $k_b = 0$  in equation (6) plotted versus the modeled flux.



**Figure 5.** Contribution to the GasEx-1998 air-sea CO<sub>2</sub> flux by breaking wave processes, obtained by subtraction of the modeled interfacial flux ( $k_b = 0$ ) from the directly measured flux.





**Figure 6.** Composite averages and standard deviations of the fluxes from three deployments of the gas flux instrumentation aboard the NOAA ship *Ronald H. Brown* plotted versus measured air-sea fugacity difference times the parameterized transfer velocity and solubility.

1998, it was determined that the bias in the fluxes due to ship motion for low wave states was approximately 2 mol m<sup>-2</sup> yr<sup>-1</sup>, and this factor was subtracted from the resulting flux estimates [McGillis *et al.*, 2001a, Figure 2b]. It is expected that there are wave state dependencies of the ship motion effect on the NDIR instrument, but the simple constant bias was assumed for this exercise. Because of the ubiquitous light winds and low fluxes encountered during this deployment, we will only report the composite averaged flux here.

[43] In Figure 6, we show the composite mean and standard deviation of the fluxes from the three deployments (GasEx-1998, JASMINE/Nauru'99, and GasEx-2001). In fact, these error estimates were computed from the difference between the measured flux and the modeled flux in order to remove the wind speed dependence from the composites. Standard errors were also computed, but the resulting plot was less than the size of the mean data circles shown in Figure 6. The means of the measured fluxes are plotted against  $\Delta f\text{CO}_2 * k * \alpha$ , which is similar in structure to equation (8). This plot shows the bias in the mean flux results which are well within the standard deviation of each sample. Note that the GasEx-2001 results show a small bias of order 1 mol m<sup>-2</sup> yr<sup>-1</sup>, and it is reasonable to expect that this bias may be explained through the arguments made earlier in section 5.1 or by a systematic bias error in the covariance measurements. A very small bias remains in the JASMINE/Nauru'99 composite, but it is encouraging that the mean flux lies very close to the 1:1 line as shown. In addition, the JASMINE/Nauru'99 results show that it is possible (albeit difficult) to make direct covariance carbon dioxide flux measurements in low flux regimes.

#### 5.4. Parameterizations of the Normalized Gas Transfer Velocity

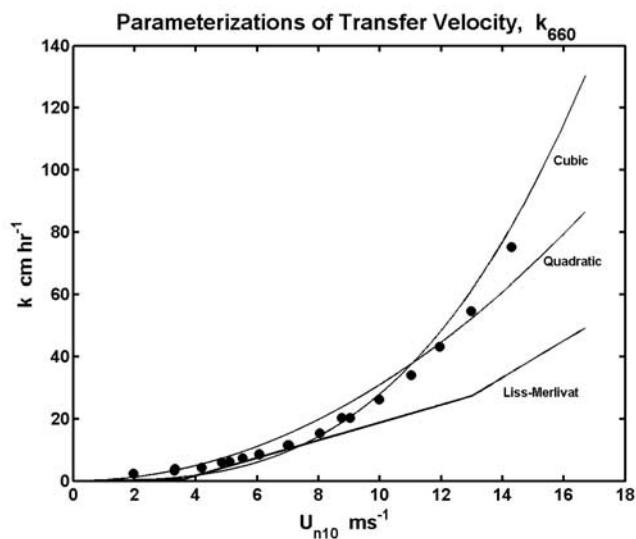
[44] As a final demonstration of the results, the wind speed bin average parameterized GasEx-1998 and GasEx-

2001 gas transfer velocities are shown in Figure 7, plotted against the 10 m neutral stability wind speed. These values of  $k$  are normalized to a Schmidt number of 660, which is a standard practice in the gas transfer literature. In addition, we have plotted some of the proposed wind speed based parameterizations of the gas transfer velocity from Wanninkhof [1992], Wanninkhof and McGillis [1999], and Liss and Merlivat [1986].

[45] As can be seen in this figure, the parameterized transfer velocities collapse onto one curve. This is not unexpected, since the Schmidt number normalization correction effectively removes differences in gas diffusivity and because the NOAA/COARE model is only mildly dependent on the difference in the observed surface net heat flux for the two data sets. This is also an indication of the relative insensitivity of the model to the averaged buoyant processes for these particular data sets. As expected, the parameterized values of  $k$  closely resemble the cubic wind speed relationship of Wanninkhof and McGillis [1999], since both results have been adjusted to the GasEx-1998 data set.

## 6. Conclusions

[46] The NOAA/COARE gas flux parameterization was run using the GasEx-1998 and GasEx-2001 experimental data sets including the warm layer and cool skin parameterizations from the COARE bulk algorithm code. Coincident directly measured fluxes from GasEx-1998 were compared to the NOAA/COARE parameterization output, and the model was tuned to this data set. This tuning process required significant adjustment (factor of 2) to the Woolf [1997] coefficient and also required that the Soloviev and Schlüssel [1994] constant be adjusted by a



**Figure 7.** Plot of the wind speed bin-averaged GasEx parameterized transfer velocities, normalized to a Schmidt number of 660, versus the 10 m neutral wind speed. Also plotted are three representative wind speed only parameterizations of transfer velocity from Wanninkhof [1992], Wanninkhof and McGillis [1999], and Liss and Merlivat [1986].

789 factor of three. Comparison of the wind speed bin-averaged  
 790 flux values from the model and the directly measured fluxes  
 791 yielded a good fit for the GasEx-1998 data, while some  
 792 significant differences were seen from the GasEx-2001 flux  
 793 results. These differences could be due to systematic bias in  
 794 the direct covariance CO<sub>2</sub> flux measurements between the  
 795 two cruises, to complications from biological processes and  
 796 surfactants, or as a result of poor model characterization of  
 797 the oceanic mixed layer processes for the two distinct  
 798 experimental regimes. There may also be some unforeseen  
 799 effect resulting from the fact that the gas transfer acts in  
 800 opposing directions (upward flux for GasEx-2001 and  
 801 downward gas flux for GasEx-1998) for the two cases. It  
 802 is seen that the GasEx-2001 results show relatively larger  
 803 interfacial fluxes, while the wave-breaking and bubble  
 804 processes are weaker.

805 [47] The most significant finding of this work is the  
 806 estimation of the relative magnitude of the gas transfer  
 807 which is due to higher wind, breaking wave processes as  
 808 opposed to interfacial exchange. This estimate was made by  
 809 comparing the GasEx-1998 measured CO<sub>2</sub> fluxes against  
 810 the NOAA/COARE model output. It was seen in Figure 5  
 811 that 80% of the flux at moderate wind speeds occurred  
 812 because of bubbles and wave breaking, which has clear  
 813 implications for accurately modeling carbon dioxide fluxes  
 814 at the higher wind speed regimes.

815 [48] There are complex processes occurring in the  
 816 oceanic surface layer, including biological activity, sur-  
 817 factants, wave breaking and bubbles, etc. It is expected  
 818 that these processes play a significant role in gas ex-  
 819 change, but considerable progress remains to be made in  
 820 order to accommodate more appropriate parameterizations  
 821 of these effects into the model. It is anticipated that  
 822 progress in all aspects of gas exchange may contribute  
 823 to the improvement of the parameterization in the future.  
 824 Significant progress on the NOAA/COARE algorithm  
 825 may also be possible through inclusion of a biological  
 826 model. Because of the significant flux which occurs at  
 827 high winds, it is also important that progress be made in  
 828 parameterizing the transfer coefficient over breaking  
 829 waves. In addition, focused process studies of gas ex-  
 830 change with coincident measurements of CO<sub>2</sub> flux and all  
 831 other relevant physical and biological processes will  
 832 continue to provide vital details which can be incorpo-  
 833 rated into the model.

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 839 are due to Robert Castle from NOAA/AOML for the maintenance and  
 840 continued development of the underway fCO<sub>2</sub> system on the *Brown*, to  
 841 Jonathan Ware for the development of the WHOI closed path atmospheric  
 842 CO<sub>2</sub> measurement system, and to the NOAA/ETL and CU/CIRES engi-  
 843 neering staff: Scott Abbott, Jesse Leach, David Costa, Sergio Pezoa, and  
 844 Brian Templeman.

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